Production of polyhydroxyalkanoates by purple phototrophic bacteria using wastewater treatment products

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Introduction

Rhodospirillum rubrum is a purple non-sulphur bacterium (PNSB) belonging to the a-proteobacteria group well-known for its metabolic versatility. This metabolic versatility leads Rs. rubrum to the forefront of the stage in the biotechnological field and as a reference organism for the study of photosynthesis and assimilation of different carbon sources. Among those carbon sources, due to their cheapness, volatile fatty acids (VFAs) are extensively studied in the biotechnology field especially in PHA production.



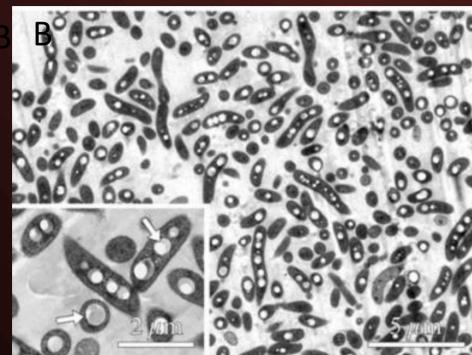




Figure 1: Transmission Electron Microscopy (TEM) pictures of Rhodospirillum rubrum in presence of different carbon sources showing PHA granules. A) succinate. B) acetate. C) butyrate

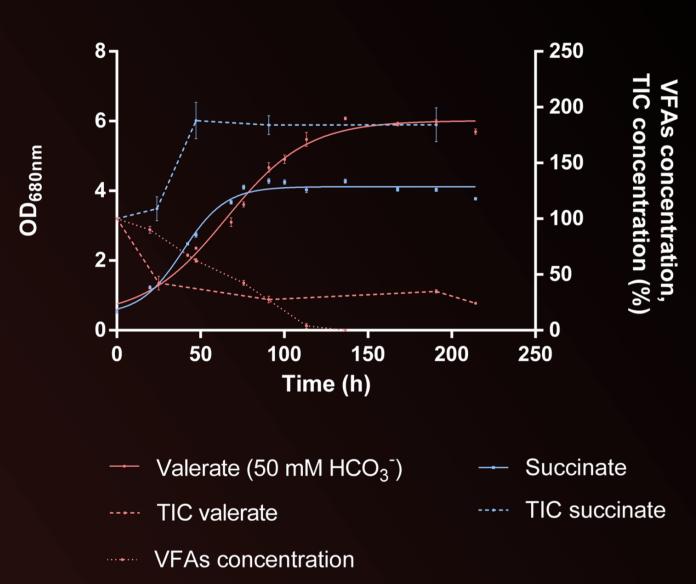
Figure 2: Metabolic pathway leading to the production of PHB and consuming a molecule of NADPH

Impact of carbon source

Redox state NAD(P)H Biomass NAD(P)+ Carbon source

Figure 3: Schematic redox yield of the assimilation of a reduced carbon source

The use of reduced carbon sources, such as VFAs, induces the production of reduced cofactors (Fig. 3). Valeric acid constitutes one of the most reduced VFAs and its assimilation by Rhodospirillum rubrum is strictly associated to the presence of HCO₃-(Fig. 4). HCO_3^- could be used as electron sink and are thus competing with PHA production. PHA quantitation revealed higher PHA content when HCO₃ are added progressively.



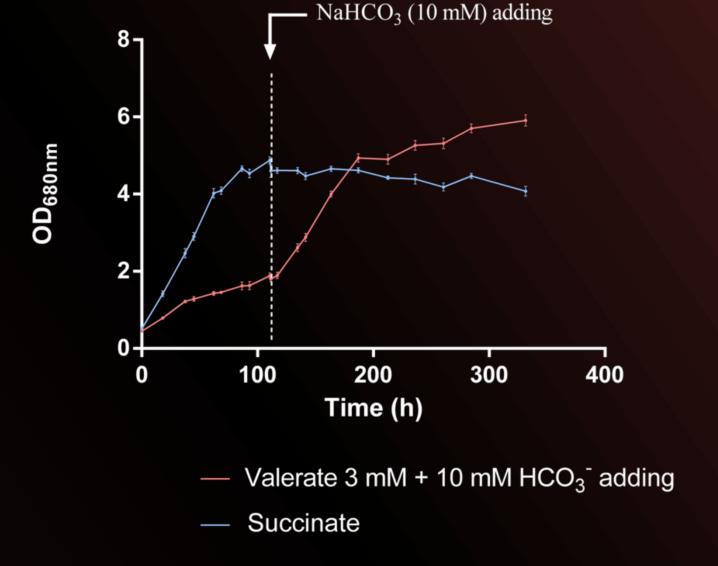
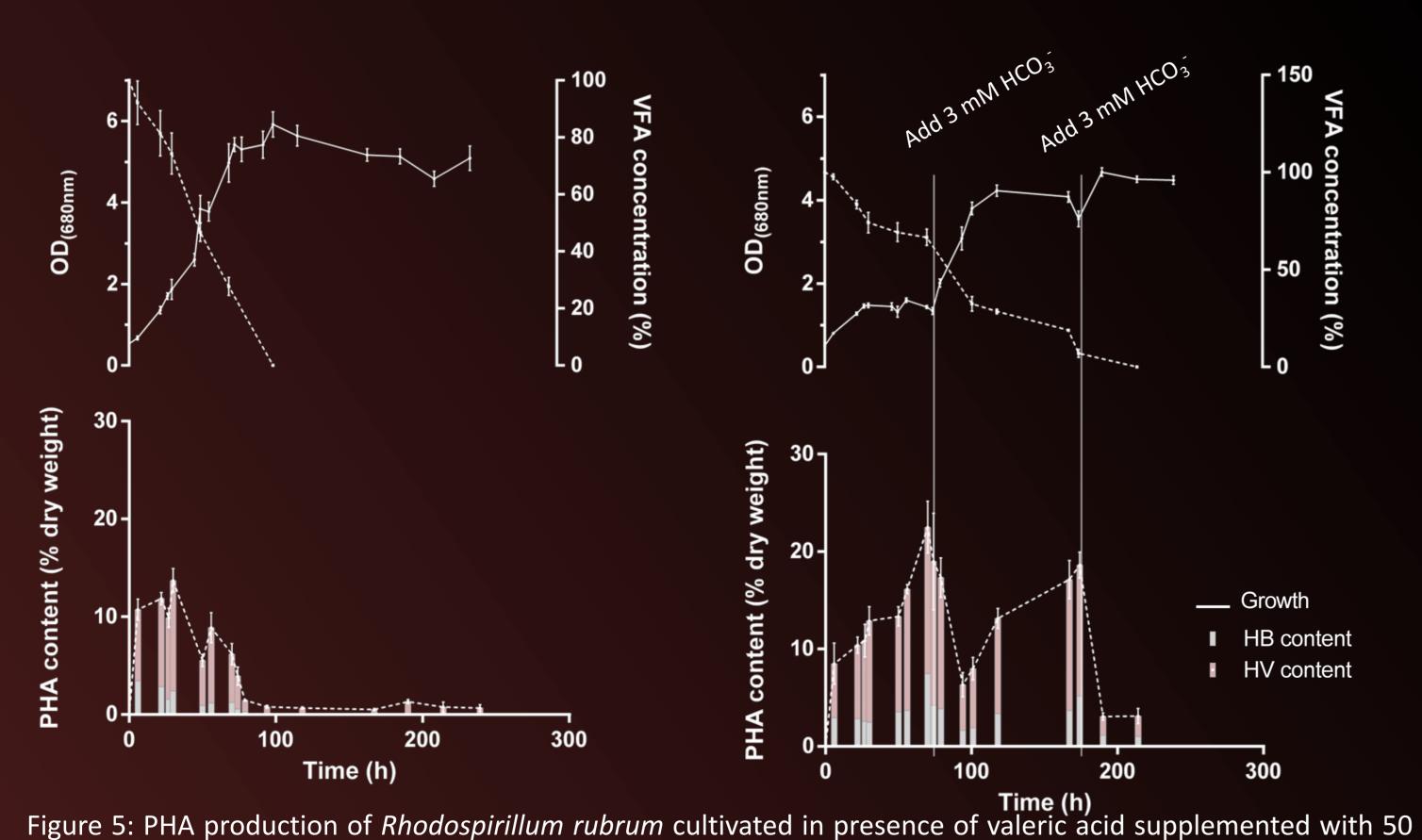


Figure 4: Rhodospirillum rubrum growth in presence of succinic and valeric acid and following of the VFAs concentration and Total Inorganic Carbon (TIC)

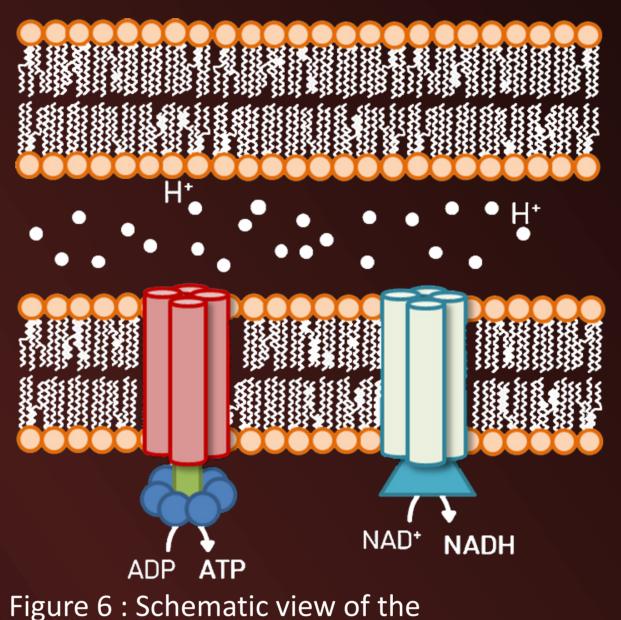
Table 1: PHA synthesis related proteins highlighted by proteomic analysis comparing valerate to succinate

Accession number	Locus Tag Identified Description		Description	<i>p</i> -value	fold change	
Q2RQI1	Rru_A2817	24	Phasin	0.0004	48.24	
Q2RP67	Rru_A3283	80	Phasin	0.0278	2.06	
Q2RNZ5	Rru_A3356	3	Polyhydroxyalkanoate depolymerase	0.0041	0.58	



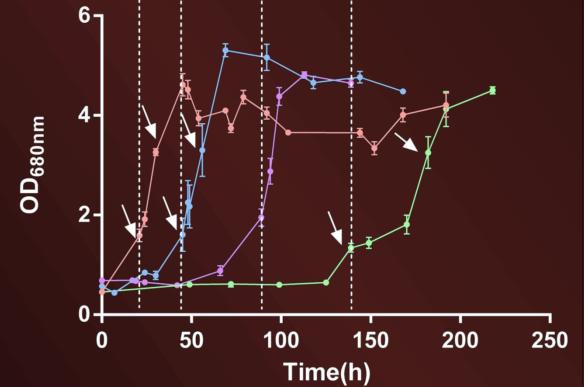
mM NaHCO₃ or with progressive adding of 3mM NaHCO₃. Dotted lines represent addition of bicarbonates ions

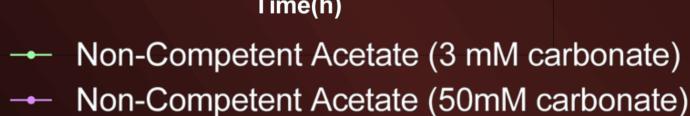
Impact of light induced stress

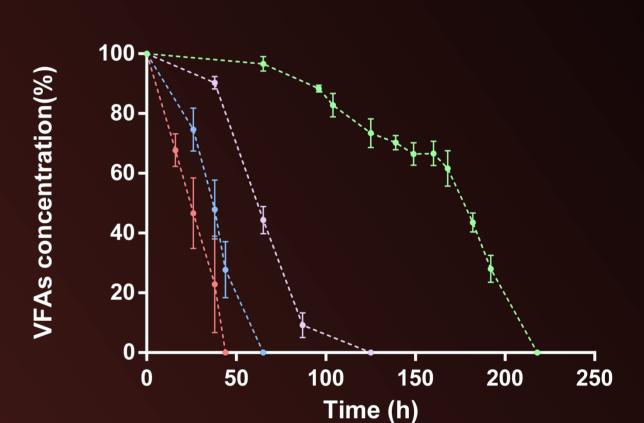


photoreduction of NAD+

High light induces intensity deregulation of the cofactor redox state via the reverse activity of the NADH dehydrogenase (Fig. 6). Our results showed that high light intensity switch induces a stop of the WT strain in presence of acetic acid as well as a stop of acetic acid assimilation (Fig. 7). analysis revealed Proteomic development of a redox stress response (Fig. 8). This was accompanied by the expected higher PHA content (Table 2).







Succinate (3mM carbonate) Compentent Acetate (3 mM carbonate)

Figure 7: Growth of Rhodospirillum rubrum WT and acetate competent strain in presence of acetic acid and response to high light intensity switch



Figure 8: Heatmap representation of differential abundance of protein considered for the analysis. Columns were separated by using Euclidian Hierarchical PRGb Forced Clustering based on conditions. Rows were separated by Unforced Euclidian hierarchical clustering. Protein Response Groups (PRGs) were established based on colour pattern and supported by Monte-Carlo simulation.

Table 2: PHA synthesis related proteins highlighted by proteomic analysis before and after light intensity increase

Accession number	Locus tag	Identified peptides	Description	Ace Comp post vs Ace Comp pre		Ace non comp post vs Ace non comp pre		Succ post vs Succ pre	
				p-value	fold change	p-value	fold change	p-value	fold change
Q2RP67	Rru_A3283	5	Phasin	0.632	1.108	0.030	0.157	0.236	0.880
Q2RNZ5	Rru_A3356	1	Polyhydroxyalkanoate depolymerase	0.008	3.279	0.010	0.267	0.061	2.404
Q2RXR4	Rru_A0276	5	Polyhydroxyalkanoate synthesis repressor PhaR	0.474	0.947	0.013	0.436	0.497	0.932

Conclusion

Our researh indicates that PHA production in Rhodosprillum rubrum is driven by intracellular redox state. The resulting strategy could be the mimicking of the carbon feast and famine process actually used in industry. The understanding of this phenomenon is mandatory to increase the yield of the PHA production and make it economically profitable.







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