

Chemical Constituents of *Crossopteryx febrifuga* (Afzel ex G. Don) Benth (*Rubiaceae*), their Phytotoxicity and Antioxidant Activities

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ABSTRACT

The aim of this work was to isolate, characterize, and evaluate the antioxidant activity as well as the phytotoxicity of the crude extracts and pure compounds obtained from *C. febrifuga*. Nine compounds were isolated from leaves, stem bark and roots crude extracts (methylene chloride/methanol: 1/1) after repeated silica gel column chromatography. corresponding to shanzhiside methyl ester (1), D-mannitol (2), a mixture of two isomeric pentacyclic triterpenes: Oleanolic acid (3a) and ursolic acid (3b), 11-methylxoside (4), 3 β -urs-12,20(30)-diene-27,28-dioic acid (5), quinovin glycoside C (6), 3-O- β -D-3'-oxo-glucopyranosyl-ursa-12,20(30)-diene-27,28-dioic acid (7) and 3 β -(α -L-rhamnopyranosyloxy)-ursa-12,20(30)-diene-27,28-dioic acid (8). The structures of these compounds were established by studying their ¹H and ¹³C NMR (1D and 2D) and Mass Spectrometry (MS) spectra. The antioxidant capacity of the crude extracts was evaluated using a recently developed electrochemical method consisting in interacting with the superoxide anion radical (O₂^{•−}) generated in situ by reduction of air molecular oxygen (O₂) in aprotic medium (anhydrous DMF). The antioxidant index is expressed as the amount of extract needed to consume 30% (AI₃₀) of electrogenerated O₂^{•−}. The lower is the antioxidant index, the more active the substrate is against O₂^{•−}. The stem bark crude extract is the most active with an AI₃₀ of 79.50 mg/L while the crude extracts of the leaves and root showed moderate activity against O₂^{•−} with AI₃₀ of 303.70 mg/L and 450.00 mg/L respectively. The phytotoxicity study was carried out using a germination test on *Lepidium sativum* seeds in the absence (0 mg/L) and presence of increasing concentrations (10 mg/L, 100 mg/L, 500 mg/L, 1000 mg/L) of the crude extracts and the isolated compound (1). The following physiological parameters of the seedlings were evaluated 72 hours after the experimentation: Germination percentage (G%), Germination Index (GI), Vigor Index (VI), Relative Root Growth (RRG%). The root extracts are the most phytotoxic on the seed germination with G% ranging from 66.66 to 100% and RRG% from 8.20 to 88.45% while the isolated compound (1) stimulated growth between 10-500 mg/L but showed moderate phytotoxic effect on seedlings above 500 mg/L.

Keywords: *Crossopteryx febrifuga*; Crude extracts; Isolation; Superoxide radical anion; Antioxidant activity; Phytotoxicity.

ABBREVIATIONS

(AI) Antioxydant Index; (AI₃₀) Antioxydant Index needed to consume 30% of O₂ in the process; (Bu₄NPF₆) Tetrabutylammonium HexafluoroPhosphate; (°C) Degree Celsius; (¹³CNMR) Carbon-13 Nuclear Magnetic Resonance; (CC) Column Chromatography; (CH₂Cl₂) Ethylene Chloride; (CM) Centimeter;

(Conc) Concentration; (COSY) Correlation-Spectroscopy; (CV) Cyclic Voltammetry; (1D) One Dimension; (2D) Two Dimensions; (D) Doublet; (dd) Doublet of doublets; (ddd) Doublet of Doublet Debouled; (DI) Insaturation Degree; (DMF) Dimethylformamide; (DMSO) Dimethylsulfoxide; (DMSO-d₆) Dimethyl sulfoxide hexadeuterated; (EtOAc) ethyl acetate; (dt) Doublet of

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triplets; (g) Gram; (G%) Germination percentage; (GI) Germination index; (GPES) General Purpose Electrochemical System; (^1H NMR) Proton Nuclear Magnetic Resonance; (H_2SO_4) Sulfuric Acid; (HMBC) Heteronuclear Multiple Bond Connectivity; (HRESI-MS) High Resolution ElectroSpray Ionization Mass Spectra; (HSQC) Heteronuclear Single Quantum Coherence; (Ip) Peak current; (Ipa) Peak anodic current; (IR) Infra-Red; (J) Coupling constant; (KBr) Potassium bromure; (Kg) kilogram; (L) Liter; (m) Multiplet; (MeOH-d_4) Methanol Tetra Deteured; (MG) Milligram; (M/Z) mass/charge; (MHz) Megahertz; (Min) Minute; (ML) Milliliter; (MM) Millimol Per Liter; (MM) Millimeter; (MS) Mass Spectra; (O) Overlap; (PH) Potential Hydrogen; (PPM) Part Per Million; (ROS) Reactive Oxygen species; (RRG%) Relative Root Growth; (S) Second; (S) Singulet; (TLC) Thin Layer Chromatography; (TMS) Tetramethylsilane; (UV) Ultraviolet; (V) volt; (V/V) Volume/Volume; (VI) Vigor Index; (δ) Chemical Shift; (μL) Microliter; (μS) Micro Siemens.

INTRODUCTION

Crossopteryx febrifuga, etymologically fighting fever, is a mono-specific genus of *Rubiaceae* family and belonging to the *Ixoroideae* subfamily [1]. The *Rubiaceae* have about 630 genera and more than 13,000 species divided into 3 subfamilies including *Rubiodeae*, *Cinchinoideae* and *Ixoroideae*, making this family one of the six largest families of Angiosperms [2]. *C. febrifuga* is a characteristic looking shrub with opposing branches that can reach 5 to 6 m in height. It gives oval and opposite leaves often reddish at their base, very fragrant white flowers grouped in clusters at the top of the twigs, its round and green fruits turn blackish when ripe. The flowering period is from May to July. This shrub grows in dry forests on the sandy soils of southern Africa [3]. It is known under vernacular names of *Ndeubeuh mann* by indigenous peoples in southern Chad and *Balembo* by Bambara peoples in Mali [4]. *Balembo* is also the name of local antitussive syrup made with *C. febrifuga* roasted fruits. This plant is used in African folk medicine to treat various diseases such as intestinal worms, cough, gout, gonorrhea, fever, diabetes and diarrheal diseases [4-7]. In southern Chad, this species is known to have many therapeutic properties, thus a decoction of root powders is taken against stomachaches. In rural areas, the same decoction is recommended to women after childbirth for its antibiotic effects. The scraping of stem bark, brown in color, is sold in local markets in the form of herbal tea, commonly called Goub, for its taste and medicinal properties while a decoction of young leaves is taken as antipyretic against fever attacks. The leaves are also used in traditional veterinary medicine as a dewormer for livestock. This species is actually threatened with extinction in Chad due to its excessive use in folk medicine and overgrazing. Chadian Government recently raised the possibility of creating a botanical garden in collaboration with the National Association of Traditional Healers in order to safeguard interesting wild plant threatened with extinction. This measure could save *C. febrifuga* in southern Chad. Some recent

phytochemical studies revealed the presence of iridoids, phenols, flavonoids, tannins, triterpenes, saponins and steroids in *C. febrifuga* extracts [8-10]. Secondary metabolites with high pharmacological potentials isolated from *Rubiaceae* justify the growing interest in the scientific investigation of the species in this family [11]. Despite the plant's many therapeutic properties recognized in ethnopharmacology, relatively few studies have been carried out in order to validate them scientifically. In this study, we evaluated for the first time the phytotoxicity of the crude extracts and isolated compound (1) on the germination and growth of *Lepidium sativum* seeds. The antioxidant capacity was measured according to a relatively new electrochemical method consisting in evaluating the capacity of the substrate to trap the superoxide anion radical ($\text{O}_2^{\bullet-}$) generated in situ by reduction of molecular oxygen (O_2) [12]. The superoxide anion radical is one of the reactive oxygen species called "ROS" whose unregulated accumulation in the human body can cause various disorders such as cancers, neurodegenerative diseases [13,14]. We also report direct isolation of a saponin (3β - $(\alpha$ -L-rhamnopyranosyl)-ursa-12,20(30)-diene-27,28-dioic acid) (8) previously obtained from *C. febrifuga* extracts by acid hydrolysis of a bisdesmosidic saponin as well as eight other known compounds.

MATERIALS AND METHODS

General experimental procedures

The masses were measured on a 1/1000 Sartorius type electronic balance. Column chromatography (CC) was carried out on a high purity grade silica gel adsorbent, pore size 60 Å, 70 mesh-230 mesh, 63-200 μm . The extracts were concentrated on a Büchi R-110 brand rotavapor. Thin Layer Chromatography (TLC) was carried out on square aluminum plates (20 \times 20 cm^2) covered with silica (0.2 mm thick) of the SIL G/UV254 Merck type, the spots were deposited with capillaries (Blaubrand® intraMark) of 5 to 20 μL and visualized under a UV lamp at 254 or 236 nm or revealed by spraying a H_2SO_4 (10%) solution followed by heating in an oven. NMR spectra were recorded on a Brüker Avance III 400 spectrometer operating at 400.13MHz for ^1H , equipped with a BBFO probe with a z-gradient coil and a GREAT 1/10 gradient unit. All experiments (^1H NMR, ^{13}C NMR and 2D as COSY, HSQC and HMBC) were carried out at 25°C. ^1H chemical shifts (δ) are given in ppm relative to the solvent residual peak and ^{13}C chemical shifts are relative to the central peak of the solvent signal. The products to be analyzed were solubilized in deuterated solvents (MeOH-d_4 or DMSO-d_6). The chemical shifts δ (ppm) were recorded with reference to tetramethylsilane (TMS, $\delta=0$). The MestReNova software was used to process the NMR spectra. The Mass Spectra (MS) are recorded on an Orbitrap Q- Extractive type apparatus (Thermo Fisher Scientific, Waltham (MA), USA) in direct infusion, in Electrospray ionization (+)/(-) modes. A JASCO 320-A type IR spectrophotometer using KBr pellets recorded the IR spectra. The cyclovoltammogram was performed on a Potentiostat Galvanostat (Autolab) Echochemie type measuring

device coupled to a computer. The data were recorded and processed by GPES Manager Software (General purpose electrochemical system, version 4.9, Echo chimie B.V). A thermostatically controlled cell containing 10 mL of 0.1 M solution (Bu_4NPF_6), maintained at $21 \pm 0.3^\circ\text{C}$ by a Fisher scientific polystat 5L/8662H type thermoregulator, in which were bathed 3 electrodes playing specific roles, namely a working electrode at glassy carbon disc (polished using a struers Dap-7 polisher fitted with a green Sci-paper, grit 1200, size: 200 mm dia, and rinsed with distilled water then dried after each use), an electrode reference (Ag/AgCl) in ethanol saturated with LiCl and a counter electrode, in the platinum wire which completes the circuit by leading the electrons through the solution to the working electrode. The germination bioassay was carried out in sterile Pétri boxes (\varnothing 90 mm) containing a layer of filter paper.

Chemicals and instruments

Commercial chemicals were used in the phytochemical study as well as in the anti-oxidant assay. The solvent systems ($\text{CH}_2\text{Cl}_2/\text{MeOH}$, 1:1) were used for all the extraction phases. n-hexane, ethylacetate and methanol were used as eluting solvents for CC. Tetrabutyl ammonium hexafluorophosphate (Bu_4NPF_6 , $M=387.43 \text{ g.mol}^{-1}$), Puriss electrochemical grade, anhydrous N, N-Dimethylformamide (DMF) used for electrochemical study were obtained from Sigma-Aldrich Switzerland. The reagents are of analytical purity and were purchased from Sigma-Aldrich. The CC tubes are of different sizes depending of the quantity of the treated sample. The extracting tanks are gloomy and hermetic to avoid any photochemical reaction as well as solvent evaporation during the extraction phases.

Plant material

Leaves, stem bark and root scraping of *C. febrifuga* were freshly harvested, in March 2014, in Béti, Department of Pendé/Logone Oriental Region, Republic of Chad, after identification by Mr Dodorom Téblé Wolwäi, a botanist at the Faculty of Science and Technics, University of Doba (Chad) and confirmed by the Yaoundé National Herbarium Voucher specimen number 41791/ HNC. These samples were transported to the laboratory and gently cleaned with distilled water to remove dust and then dried for two weeks at room temperature out of direct sunlight.

Extracts preparation

After preliminary stages of drying and manual powdering the plant material; 1.0 kg of powder from the leaves; 1.0 kg of powder from the stem bark and 1.0 kg of powder from the root were separately extracted by maceration, at room temperature, for 48 hours with 6 L of methylene chloride/methanol (1/1, v/v) system. Maceration is a liquid-solid extraction method that takes place at room temperature and has the advantage of preserving heat sensitive molecules before any intentional modification of their structures in organic synthesis. This operation was repeated twice in order to optimize the extraction. The different resulting solutions were filtered using Whatman paper and then

concentrated under reduced pressure at 55°C on a rotavapor to give successively 80 g of a soft green brown paste for the leaves, 80 g of an amorphous solid of chocolate color for the root and 100 g of raw extract of a creamy consistency and red brown in color for the stem bark. These crude extracts were stored at 4°C until the next stage of the study.

Column chromatography and isolation

Samples of 50 g of crude extracts were subjected separately to liquid silica gel column chromatography (pore size 60\AA , 70 mesh-230 mesh, $63\text{--}200\mu\text{m}$, 300 g) and eluted with n-hexane/ethylacetate and ethylacetate/methanol systems in increasing polarity (100:0 \rightarrow 0:100). Fractions of 200 mL each collected at the outlet of the chromatographic column were concentrated using a rotavapor. This operation yielded a total of 218 fractions for the stem bark crude extracts, 340 fractions for the leaves crude extracts and 400 fractions for the root crude extracts and then gathered into major sub-fractions according to their thin layer chromatographic profiles. Successive purifications of some UV absorbing sub-fractions afford nine compounds identified using spectroscopic NMR, Mass Spectrometry methods and corresponding to the following known compounds:

Leaves crude extracts

Four compounds were isolated from this extract: Shanzhiside methyl ester (1), a white powder (250.20 mg) precipitated in the $\text{EtOAc}/\text{CH}_2\text{Cl}_2$ 1:1 system; D-mannitol (2), white powder (19.7 mg, $\text{EtOAc}/\text{CH}_2\text{Cl}_2$ 1:1); a mixture of oleanolic acid (3a) and ursolic acid (3b); white powder (31.71 mg, $\text{n-C}_6\text{H}_{14}/\text{EtOAc}$ 7.5:2.5).

Root crude extracts

Three compounds including 11-methylxoside (4), yellowish amorphous powder (7.52 mg, EtOAc/MeOH 9:1); 3β -urs-12,20(30)-diene-27,28-dioic acid (5), white powder (5.23 mg, $\text{n-C}_6\text{H}_{14}/\text{EtOAc}$ 3:2) and quinovinglycoside C (6), white amorphous powder (8.63 mg, $\text{n-C}_6\text{H}_{14}/\text{EtOAc}$ 1:9).

Stem bark crude extracts

Two compounds including 3-O- β -D-3'-oxo-glucopyranosyl-ursa-12,20(30)-diene-27, 28-dioic acid (7), brownish powder (11.80 mg, $\text{n-C}_6\text{H}_{14}/\text{EtOAc}$ 1:1) and 3β -(α -L-rhamno- pyranosyl)-ursa-12,20(30)-diene-27,28-dioic acid (8), beige powder (18.0 mg, EtOAc/MeOH 9:1) were isolated.

Phytotoxicity test

We evaluated the phytotoxicity of crude extracts and the isolated compound (1) on the germination and growth of *Lepidium sativum* seeds, a vegetable plant of *Brassicaceae* family. The study was performed according to the protocol described by Mihaela with a slight modification [15]. For each sample, series of concentrations of 10, 100, 500 and 1000 mg/L were prepared using distilled water. The solutions obtained were further homogenized in an ultrasonic bath. Negative control was made with distilled water containing no substrate (0 mg/L). Ten (10) seeds of regular

contours and similar sizes, selected after a visual examination, are inoculated with 5ml of each solution previously prepared in sterile Petri dishes where filter papers were previously placed, serving as the germination surface. The seeds were distributed over the entire surface of the box to prevent any hindrance during the germination period. This operation was repeated 3 times for each concentration. Before the experimentation, physico chemical parameters of solutions such as pH and conductivity ($\mu\text{S}/\text{cm}$) were measured. After 3 days of incubation at room temperature and protected from light, the following physiological parameters of the seedling were measured: germination percentage (G%), germination index (GI), relative root growth (RRG%) and vigor index (VI) expressed as follows [16,17].

$$G\% = \frac{\text{Nbr of germinated seeds in the sample}}{\text{Nbr of germinated seeds in the control}} \times 100$$

$$RRG\% = \frac{\text{Mean of root length in the sample}}{\text{Mean of root length in the control}} \times 100; \quad GI = \frac{G\%}{RRG\%} \times 100$$

$$VI = \frac{\text{Mean of root length}}{\text{Mean of shoot length}} \times G\%$$

In the germination process, we considered as germinated seeds with root lengths ≥ 1 mm. Comparisons with the control (0 mg/L) allowed us to know the allelopathic response of the seedling to the external stress. Allelopathy is any direct or indirect effect, positive or negative, of a plant on another through the production of chemical compounds released into the environment [18]. The allelopathic activity of Brassicaceae family is attributed to glucosinolates, a group of sulfur containing carbohydrate compounds common in this family [19,20].

Determination of the antioxidant activity of the crude extracts

0.1 M solutions (Bu_4NPF_6) were prepared by dissolving 0.387 g of Bu_4NPF_6 (support electrolyte to conduct the current) in 10 mL of anhydrous DMF (99.8%) and saturated with dry air for 10 min. Under the operating conditions, the solubility of oxygen in the air is approximately $S(\text{O}_2) = 0.94 \cdot 10^{-3} \text{ M}$. First, the molecular oxygen (O_2) reduction in cyclic voltammogram was recorded, in the absence of any antioxidant substrate, at a scan rate of $0.1 \text{ V} \cdot \text{s}^{-1}$. The initial potential was set at 0 V vs Ag/AgCl. The sweep potential was reversed at -0.9 V relative to Ag/AgCl. The presence of the radical ($\text{O}_2^{\bullet-}$) in the medium is easily detected by measuring its anodic oxidation current (I_{pa}) during reverse scanning [12]. We then prepared 5 mL of stock solutions (10 g.L⁻¹) each by dissolving crude extracts (root, stem bark, leaves) in anhydrous DMF. Gradual volumes of these freshly prepared stock solutions were added to the 10 mL of 0.1 M solution (Bu_4NPF_6) in the thermostatically controlled cell to obtain a range of antioxidant substrate concentrations from 0 mg.L⁻¹ to 700 mg.L⁻¹. After each addition, the cyclic voltammogram of the progressive reduction of the superoxide radical was recorded. The consumption of $\text{O}_2^{\bullet-}$ is correlated with the concentration of crude extracts gradually added in the electrochemical cell, thus reflecting their antioxidant capacity. This antioxidant capacity is estimated by the Antioxidant Index (AI_{30}) defined as the concentration of extract required to consume 30% of

electrogenated superoxide anion radical. The smaller the AI_{30} , the more active the extract is against the super-oxide anion radical [21].

RESULTS AND DISCUSSION

Structures of the isolated compounds

After a column chromatographic fractionation of the crude extracts followed by successive purifications on sephadex LH20, the following nine known compounds, belonging to four organic compounds groups (iridoids, triterpenes, saponins and carbohydrates) were isolated. Their structures were identified by 1D (¹H, Mitragyna Stipulosa [29], 3-O- β -D-3'-oxoglucopyranosylC, DEPT135), 2D (COSY, HSQC, HMBC) NMR spectra and by comparison with the literature data: Shanzhiside methyl ester (1) [22], D-mannitol (2), a primary metabolite common to several species [23], a mixture of two isomeric pentacyclic triterpenes: Oleanolic acid (3a) and ursolic acid (3b) [24-26], 11-methylxoside (4) [27], 3 β -urs-12,20-diene-27,28-dioic acid (5) [28], quiovins glycoside C (6), an antivenom saponoside previously isolated from extracts of Mitragyna Stipulosa [29], 3-O- β -D-3'-oxoglucopyranosyl-ursa-12,20(30)-diene-27,28-dioic acid (7) [8] and 3- β -(α -L-rhamnopyranosyloxy)-ursa-12,20(30)-diene-27,28-dioic acid (8) isolated for the first time from extracts of *C. febrifuga* (Figure 1).

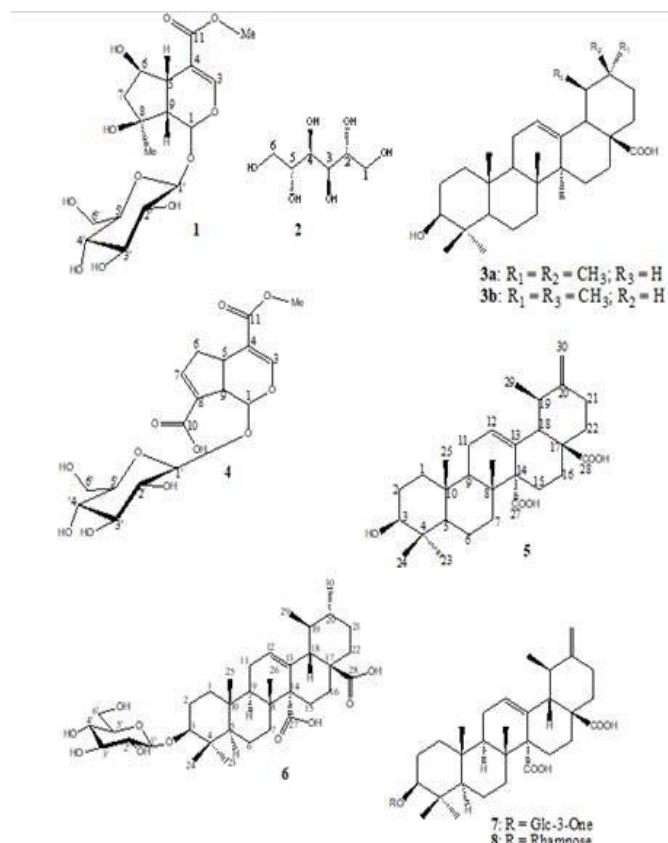


Figure 1: Chemical structures of the isolated compounds.

General informations on the isolated compounds

Compound 1 (Shanzhiside méthyl ester) white powder (ethyl acetate/methanol 1:1), (HRESI-MS) (+): m/z 429.13673 [$\text{M}+\text{Na}$]⁺ (calculated for $\text{C}_{17}\text{H}_{26}\text{O}_{11}\text{Na}$: 429.1367) corresponding to the formula $\text{C}_{17}\text{H}_{26}\text{O}_{11}$, DI=5, ¹H NMR (δ ppm, DMSO-d₆, 400MHz):

7.33 (d, $J=1.25\text{Hz}$, H-3), 5.50 (d, $J=2.3\text{Hz}$, H-1), 4.45 (d, $J=7.9\text{Hz}$, H-1'), 3.92 (m, H-6), 3.70 (ddd, $J=11.9\text{Hz}$, $J=6.4\text{Hz}$, $J=1.9\text{Hz}$, Ha-6'), 3.65 (s, OMe-12), 3.44 (dt, $J=11.9\text{Hz}$, $J=6.0\text{Hz}$, Hb-6'), 3.15(o) m, H-5', 3.12 (o) m, H-3', 3.04 (ddd, $J=9.4\text{Hz}$, $J=8.7\text{Hz}$, $J=5.3\text{Hz}$, H-4'), 2.93 (ddd, $J=8.7\text{Hz}$, $J=8.0\text{Hz}$, $J=5.4\text{Hz}$, C-2'), 2.81 (dd, $J=9.9\text{Hz}$, $J=2.5\text{Hz}$, H-5), 2.45 (dd, $J=9.8\text{Hz}$, $J=2.3\text{Hz}$, H-9), 1.83 (dd, $J=13.1\text{Hz}$, $J=6.6\text{Hz}$, Ha-7), 1.63 (dd, $J=13.1\text{Hz}$, $J=6.4\text{Hz}$, Hb-7), 1.11 (s, H-10), ^{13}C NMR (100MHz): 167.19 (C-11), 151.00(C-3), 109.6(C-4), 98.0(C-1'), 92.8(C-1), 77.33(C-5'), 76.99(C-8), 76.7(C-3'), 74.9(C-6), 73.1(C-2'), 70.02(C-4'), 61.09(C-6'), 51.25(OMe-12), 49.99(C-9), 49.18(C-7), 40.29(C-5), 24.7(C-10).

Compound 2 (D-mannitol) white Powder (ethyl acetate/methanol 1:1), ^1H NMR (δppm , DMSO- d_6 , 400MHz): 4.41(d, $J=5.6\text{Hz}$, OH-1/6), 4.34(t, $J=5.6\text{Hz}$, OH-2/5), 4.13 (d, $J=7.0\text{Hz}$, OH-3/4), 3.61(ddd, $J=10.7\text{Hz}$, $J=5.6\text{Hz}$, $J=3.4\text{Hz}$, Ha-1/6), 3.53 (t, $J=7.6\text{Hz}$, H-3/4), 3.45(ddd, $J=10.7\text{Hz}$, $J=5.6\text{Hz}$, $J=3.5\text{Hz}$, H-2/5), 3.39(dd, $J=10.7\text{Hz}$, $J=5.7\text{Hz}$, Hb-1/6), ^{13}C NMR (100MHz): 71.35(C-2/5), 69.71(C-3/4), 63.90(C-1/6).

Compounds 3a (oleanolic acid): White powder (hexane/ethyl acetate 7.5:2.5), ESI- m/z 455 [M-H] $^-$ compatible with $\text{C}_{30}\text{H}_{48}\text{O}_3$, DI=7, IR (KBr): $\nu_{\text{max}}=3,300\text{cm}^{-1}$ (-OH), $3,000\text{cm}^{-1}$ (Csp 3 -H), $1,700\text{cm}^{-1}$ (C=O), ^{13}C NMR (125MHz, DMSO- d_6): 178.2 (C-28), 143.8 (C-13), 121.5 (C-12), 76.8 (C-3), 55.9 (C-5), 48.2 (C-9), 46.7 (C-17), 46.6 (C-19), 42.2 (C-14), 42.1 (C-18), 39.4 (C-4), 39.8 (C-8), 39.1(C-1), 37.4 (C-10), 34.3 (C-21), 33.4 (C-7/29), 33.2 (C-22), 31.0 (C-20), 28.8 (C-23), 28.4 (C-15), 28.0 (C-2), 26.22 (C-27), 23.8 (C-11/16/30), 18.8 (C-6), 17.6 (C-26), 16.5 (C-24), 15.6 (C-25).

Compounds 3b (ursolic acid): white powder (hexane/ethyl acetate 7.5:2.5), ESI- m/z 455 [M-H] $^-$ compatible with $\text{C}_{30}\text{H}_{48}\text{O}_3$, DI=7, IR (KBr): $\nu_{\text{max}}=3,300\text{cm}^{-1}$ (-OH), $3,000\text{cm}^{-1}$ (Csp 3 -H), $1,700\text{cm}^{-1}$ (C=O), ^{13}C NMR (125MHz, DMSO- d_6): 178.6 (C-28), 138.2 (C-13), 124.5 (C-12), 76.8 (C-3), 55.9 (C-5), 53.6 (C-18), 48.2 (C-9), 48.1 (C-17), 42.7 (C-14), 40.1 (C-8), 39.7 (C-4), 39.3 (C-1), 39.5 (C-19), 39.4 (C-20), 37.6 (C-10), 37.4 (C-22), 33.8 (C-7), 31.1 (C-21), 28.9 (C-15), 28.8 (C-23), 28.3 (C-2), 25.0 (C-16), 24.0 (C-27), 23.8 (C-11), 21.4 (C-30), 18.9(C-6), 17.5 (C-26/29), 16.5 (C-24), 15.7 (C-25).

Compound 4 (11-Methylxoside) yellowish amorphous powder (ethylacetate/methanol 9:1), High Resolution ESI Mass spectrum (HRESI-MS) (+)/(-), ESI: m/z 425 [M+Na] $^+$ and ESI- m/z 401 [M-H] $^-$ compatible with the formula $\text{C}_{17}\text{H}_{22}\text{O}_{11}$, ^1H NMR (δppm CH $_3$ OH- d_4 , 400MHz): 7.49(d, $J=1.1\text{Hz}$, H-3), 6.90 (dt, $J=2.5\text{Hz}$, $J=1.6\text{Hz}$, H-7), 5.69(d, $J=5.0\text{Hz}$, H-1), 4.62(d, $J=7.9\text{Hz}$, H-1'), 3.86(m, Ha-6'), 3.71(s, OMe-12), 3.68(m, Hb-6'), 3.36(m, H-3'), 3.33(m, H-5), 3.30(m, H-5'), 3.29(m, H-4'), 3.22(m, H-9), 3.19(dd, $J=8.9\text{Hz}$, $J=7.9\text{Hz}$, H-2'), 2.93(dt, $J=18.7\text{Hz}$, $J=8.0\text{Hz}$, $J=2.3\text{Hz}$, Ha-6) and 2.43(m, Hb-6), ^{13}C NMR (100MHz): 169.2(C-11), 168.0(C-10), 153.5(C-3), 147.5(C-7), 136.2(C-8), 112.4(C-4), 100.4 (C-1'), 96.3(C-1), 78.2(C-5'), 77.8(C-3'), 74.6(C-2'), 71.4(C-4'), 62.6(C-6'), 51.7 (OMe-12), 47.4(C-9), 40.2(C-6) and 35.0(C-5).

Compound 5 (3 β -urs-12,20 (30) -diene-27,28-dioic acid) white powder (hexane/ethyl acetate 6:4), (HRESI-MS) (+)/(-), ESI $^+$ m/z [M+Na] $^+$ 507.30809 (calculated for $\text{C}_{30}\text{H}_{44}\text{O}_5\text{Na}$: 507.3082) and ESI-, m/z [M-H] $^-$ 483.3116 (calculated for $\text{C}_{30}\text{H}_{43}\text{O}_5$) compatible with the formula $\text{C}_{30}\text{H}_{44}\text{O}_5$ (DI=9), ^1H NMR (δppm , DMSO- d_6 , 400MHz): 5.50 (m, H-12), 4.67 (brs, Hb-30), 4.59 (brs, Ha-30), 4.32 (brs, OH-3), 2.95 (brs t, $J=7.2\text{Hz}$, H-3), 2.22 (d, $J=11.7\text{Hz}$, H-18), 2.18 (o, H-9), 2.12 (o, 2H-21), 2.05 (o, Hb-16), 1.98 (o, Ha-15), 1.9 (o, Hb-11), 1.84 (o, H-19), 1.8 (o, Ha-11), 1.70 (o, Hb-15), 1.69 (o, Ha-22), 1.63 (o, Ha-16), 1.6 (o, 2H-1), 1.6 (o, Hb-7), 1.55 (o, Ha-7), 1.46 (o, Hb-22), 1.45 (m, Ha-6), 1.45 (o, 2H-2), 1.25 (o, Hb-6), 0.95 (d, $J=6.3\text{Hz}$, H-29), 0.87 (s, 3H-25), 0.86 (s, 3H-23), 0.79 (s, 3H-26), 0.66 (s, 3H-24), 0.58 (d, $J=10.9\text{Hz}$, H-5), ^{13}C NMR (100MHz): 177.7(C-28), 176.0(C-27), 152.5(C-20), 132.5(C-13), 128.3(C-12), 105.8(C-30), 77.0(C-3), 55.7(C-18), 55.4(C-14), 55.0(C-5), 47.4(C-17), 46.12(C-9), 39.0(C-8), 38.4(C-1), 38.4(C-4), 38.1(C-22), 36.5(C-10), 36.3(C-7), 35.0(C-19), 31.5(C-21), 28.34(C-23), 27.0(C-2), 25.15(C-16), 24.1(C-15), 22.5(C-11), 18.2(C-26), 18.0(C-6), 16.1(C-24), 16.0(C-25), 15.99(C-29).

Compound 6 (quinovin glycoside C): White amorphous powder (hexane/ethyl acetate 1:9), (HRESI-MS)(+)/(−), ESI $^+$ m/z [M+Na] $^+$ 671, ESI- m/z [M-CO $_2$ -H] $^-$ 603 compatible with the formula $\text{C}_{36}\text{H}_{56}\text{O}_{11}$, DI=9, ^1H NMR (δppm , DMSO- d_6 , 400MHz): 5.49 (m, H-12), 4.14 (d, $J=7.8\text{Hz}$, H-1'), 4.02 (m, H-4'), 3.64 (dd, $J=11.0\text{Hz}$, $J=3.4\text{Hz}$, Ha-6'), 3.41 (o, Hb-6'), 3.1 (t, $J=8.6\text{Hz}$, H-3'), 3.06 (m, H-5'), 2.99 (dd, $J=11.8\text{Hz}$, $J=4.6\text{Hz}$, H-3), 2.94 (dt, $J=7.8\text{Hz}$, $J=4.6\text{Hz}$, H-2'), 2.14 (o, H-9/18), 1.92 (o, Ha-16), 1.90 (o, Ha-15), 1.87 (o, Ha-11), 1.80 (o, Ha-2), 1.78 (o, Hb-11), 1.58 (o, Ha-1), 1.57 (o, Ha-7), 1.56 (o, Hb-16), 1.54 (o, Hb-2), 1.53 (o, Hb-15), 1.50 (o, 2H-22), 1.44 (o, Ha-6), 1.39 (o, 2H-21), 1.23 (o, Hb-6), 0.94 (s, 3H-23), 0.89 (o, H-19/20), 0.87 (d, $J=4.6\text{Hz}$, 3H-30), 0.88 (s, 3H-26), 0.86 (o, Hb-1/7), 0.81 (d, $J=5.3\text{Hz}$, 3H-29), 0.78 (s, 3H-25), 0.73 (s, 3H-24), 0.6 (d, $J=11.4\text{Hz}$, H-5), ^{13}C NMR (100MHz): 178.3 (C-28), 176.6 (C-27), 132.6 (C-13), 127.9 (C-12), 105.3 (C-1'), 87.9 (C-3), 76.9(C-3'), 76.6 (C-5'), 74.0(C-2'), 70.23(C-4'), 61.26 (C-6'), 55.34 (C-14), 55.26 (C-5), 53.7 (C-18), 47.3 (C-17), 46.2 (C-9), 39.0 (C-10), 38.7 (C-4), 38.6 (C-1), 38.5 (C-20), 36.7 (C-7), 36.5 (C-19), 36.2 (C-8), 35.9 (C-22), 29.8 (C-21), 27.7 (C-23), 25.6 (C-2), 25.0 (C-15), 24.1 (C-16), 22.4 (C-11), 21.2 (C-30), 18.2 (C-25), 17.9 (C-6), 17.6 (C-29), 16.6 (C-24), 16.1 (C-26).

Compound 7 (3-O- β -D-3'-oxo-glucopyranosyl-ursa-12,20(30)- diene-27,28-dioic acid): Brownish powder (hexane/ethyl acetate 1:1), (HRESI-MS)(-): m/z 643.3492 [M-H] $^-$ (calculated for $\text{C}_{36}\text{H}_{51}\text{O}_{10}$) compatible with $\text{C}_{36}\text{H}_{52}\text{O}_{10}$, ^1H NMR (δppm , CH $_3$ OH- d_4 , 400MHz): 5.63 (m, H-12), 4.68 (s, Hb-30), 4.64 (s, Ha-30), 4.40 (d, $J=7.8\text{Hz}$, H-1'), 4.22 (dd, $J=10.1\text{Hz}$, $J=1.7\text{Hz}$, H-4'), 4.11 (dd, $J=7.8\text{Hz}$, $J=1.7\text{Hz}$, H-2'), 3.92 (dd, $J=12.0\text{Hz}$, $J=2.0\text{Hz}$, Hb-6'), 3.79 (dd, $J=12.0\text{Hz}$, $J=4.6\text{Hz}$, Ha-6'), 3.29 (ddd, $J=10.1\text{Hz}$, $J=4.6\text{Hz}$, $J=2.0\text{Hz}$, H-5'), 3.21 (dd, $J=11.5\text{Hz}$, $J=4.5\text{Hz}$, H-3), 2.32 (d, $J=11.8\text{Hz}$, H-18), 2.26 (o, H-9), 2.25 (o, Ha -16), 2.25 (o, Hb-21),

2.19 (o, Ha-21), 2.1 (o, Ha-15), 1.98 (o, H-19), 1.97 (o, Hb -11), 1.91 (o, Ha -11), 1.90 (o, Hb-2), 1.80 (o, Ha-7), 1.78 (o, Ha-2), 1.77 (o, Hb-16), 1.75 (o, Hb-15), 1.70 (o, 2H-1), 1.67 (o, 2H-22), 1.55 (o, Ha-6), 1.55 (o, Hb-7), 1.37 (o, Hb-6), 1.04 (d, $J=6.4\text{Hz}$, 3H-29), 1.02 (s, 3H-23), 0.99 (s, 3H-25), 0.90 (s, 3H-26), 0.85 (s, 3H-24), 0.77 (m, H-5), ^{13}C NMR (100MHz): 207.55 (C-3'), 180.8 (C-28), 178.8 (C-27), 154.1 (C-20), 133.6 (C-13), 131.08 (C-12), 107.9 (C-1'), 106.2 (C-30), 91.1 (C-3), 78.9 (C-2'), 78.0 (C-5'), 73.8 (C-4'), 62.6 (C-6'), 57.7 (C-18), 57.3 (C-14), 56.9 (C-5), 49.6 (C-17), 48.0 (C-9), 40.7 (C-8), 40.1 (C-4), 40.0 (C-7), 39.9 (C-1), 38.0 (C-22), 37.9 (C-10), 36.7 (C-19), 33.0 (C-21), 28.4 (C-23), 27.0 (C-2), 26.6 (C-16), 25.8 (C-15), 23.9 (C-11), 19.3 (C-6), 19.0 (C-26), 17.0 (C-24), 16.9 (C-25), 16.6 (C-29).

Compound 8 (3- β -(α -L-rhamnopyranosyloxy)-ursa-12,20(30)-diene-27,28-dioic acid): Beige powder (ethyl acetate/methanol 9:1), (HRESI-MS)(+): m/z 653.82 $[\text{M}+\text{Na}]^+$ calculated for $\text{C}_{36}\text{H}_{54}\text{O}_9$, $\text{DI}=10$, ^1H NMR (δ ppm, DMSO- d_6 , 400MHz): 5.53 (m, H-12), 4.59 (brs, Ha-30); 4.68 (brs, Hb-30), 4.15 (d, $J=7.7\text{Hz}$, H-1'), 3.15 (m, H-5'), 3.07 (dt, $J=8.9\text{Hz}$, $J=4.5\text{Hz}$, H-3'), 2.98 (d, o, H-3), 2.97 (m, H-2'), 2.78 (dt, $J=8.9\text{Hz}$, $J=5.4\text{Hz}$, H-4'), 2.23 (d, $J=11.8\text{Hz}$, H-18), 2.16 (o, H-9), 2.14 (o, 2H-21), 1.99 (o, Ha-15), 1.90 (o, Ha-11), 1.84 (o, H-19), 1.81 (o, Hb-11), 1.75 (o, Hb-2), 1.71 (o, Ha-22), 1.66 (o, 2H-16), 1.61 (o, Hb-15), 1.61 (o, 2H-1), 1.55 (o, Ha-2), 1.55 (o, 2H-7), 1.48 (o, Hb-22), 1.46 (o, Ha-6), 1.25 (o, Hb-6), 1.13 (d, $J=6.1\text{Hz}$, 3H-6'), 0.96 (d, $J=6.5\text{Hz}$, 3H-29), 0.94 (s, 3H-23), 0.89 (s, 3H-25), 0.79 (s, 3H-26), 0.74 (s, 3H-24), 0.61 (o, H-5), ^{13}C NMR (100MHz): 177.7 (C-28), 176.1 (C-27), 152.3 (C-20), 132.3 (C-13), 128.5 (C-12), 105.9 (C-30), 105.2 (C-1'), 87.9 (C-3), 76.6 (C-3'), 75.4 (C-4'), 74.2 (C-2'), 71.2 (C-5'), 55.7 (C-18), 55.3 (C-14), 55.3 (C-5), 47.4 (C-17), 46.16 (C-9), 39.06 (C-8), 38.7 (C-4), 38.55 (C-1), 38.1 (C-22), 36.2 (C-10), 36.2 (C-7), 35.1 (C-19), 31.2 (C-21), 27.7 (C-23), 25.8 (C-2), 25.1 (C-16), 24.1 (C-15), 22.4 (C-11), 18.1 (C-6'), 18.1 (C-26), 17.8 (C-6), 16.5 (C-24), 16.1 (C-25), 16.0 (C-29).

Antioxidant activity

Stem bark, Roots and Leaves crude extracts: This method of measuring antioxidant capacity of a substrate has been validated with Carine's work that has been used to measure the flavonoids antioxidant capacity [12]. Our results clearly showed that the extracts tested, using this method, showed an antioxidant activity which is quantified for a better assessment by calculating the Antioxidant Index AI_{30} ($(\text{Ipa}-\text{Ipa}^\circ)/\text{Ipa}^\circ=0.30$). The stem bark extract is the most active with a lower AI_{30} of 79.50 mg/L, meaning that a substrate concentration of 79.50 mg/L is required to consume 30% of electrogenerated $\text{O}_2^{\bullet-}$ while leaves and root crude extracts showed moderate activities with AI_{30} of 303.70 mg/L and 450.00 mg/L, respectively. Generally in Chadian folk medicine, the stem bark is the most used part of the plant, then powders of stem bark is used as a herbal tea locally called Goub, as well as a hemostatic agent for bloody wounds. From the stem bark extracts, we isolated two saponins, including 3-O- β -D-3'-oxo-glucopyranosyl-ursa-12,20(30)-diene-27,28-dioic acid known for its

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moderate antibacterial activity [8] and 3- β -(α -L-rhamnopyranosyloxy)-ursa-12,20(30)-diene-27,28-dioic acid isolated for the first time from extracts of *C. febrifuga* (Figures 2-5).

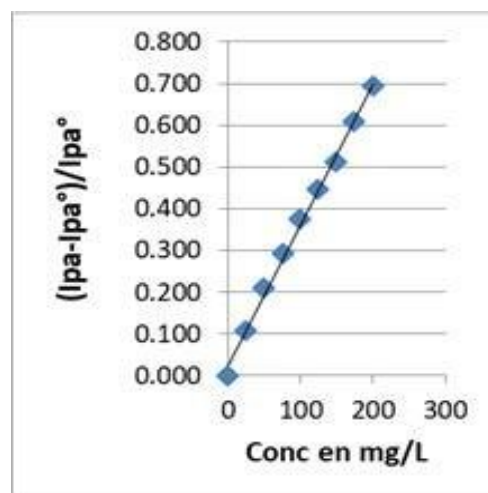


Figure 2: Linear response of $\text{O}_2^{\bullet-}$ consumption depending on the stem bark extract concentration.

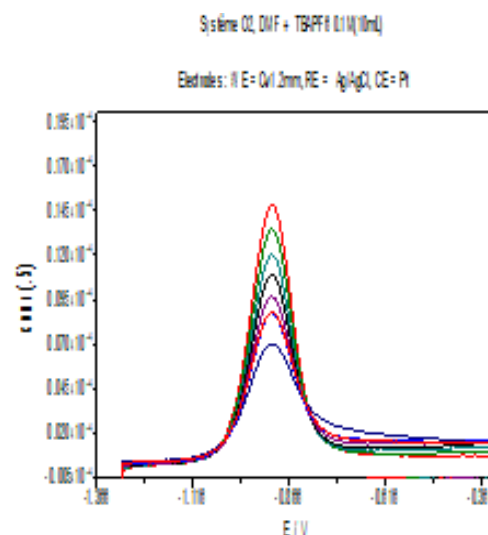


Figure 3: Cyclic voltammograms of O_2 reduction in the absence and presence of increasing concentrations of stem bark extract at a stable glassy carbon disc electrode in 0.1M (Bu_4NPF_6) solution. 0.1 V / s scan rate.

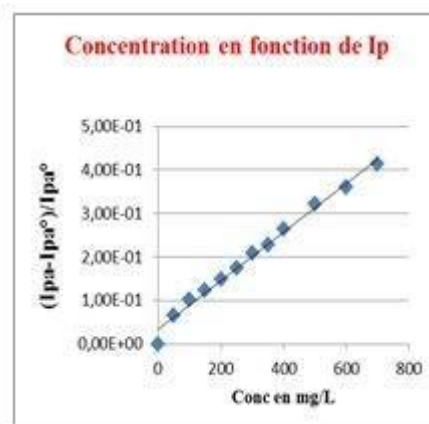


Figure 4: Linear response of O_2 consumption as a function of root extract concentration

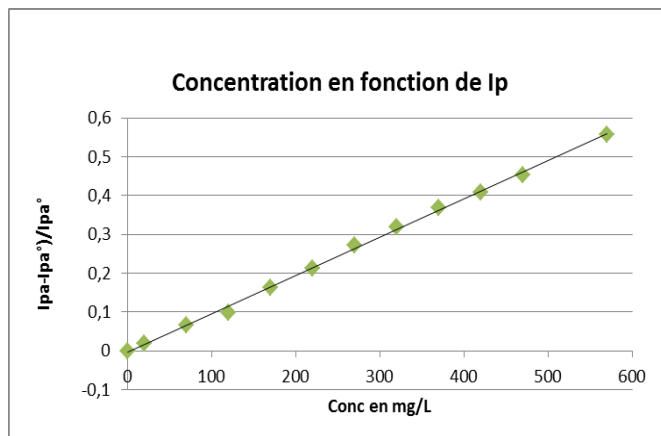


Figure 5: Linear response of O₂ consumption as a function of leaves extract

Phytotoxicity

Shanzhiside methyl ester (1) and roots crude extracts: Our results revealed that all the tested crude extracts and compound (1) exhibited phytotoxicity on the germination and growth of seeds at varying degrees. Most plants contain inhibiting substances on the seed germination and the degree of inhibition is proportional to the concentration [30]. We found that at a certain concentration, mostly relatively low (10 and 100 mg/L), the extracts have a stimulating effect on growth but become toxic or phytotoxic at high concentrations. Some compounds inhibit seed germination at 50 ppm but stimulate growth below 1ppm [30]. The root extracts are more toxic against seed germination with the germination percentage (G %) ranging from 66.66 to 100% and the relative root growth (RRG %) from 8.20 to 88.45%. From 500 mg/L, the seedlings are stunted with average root and stem lengths of 3.36 and 4.01mm respectively, whereas these values are respectively 41.05 and 13.66 mm for the control (in the absence of extract). In addition to the physiological parameters reduced between 500 and 1000 mg/L, yellowing of the foliage is observed, reflecting the external stress undergone by the seedling. Compound (1), belonging to the iridoids class, stimulated growth between 10-500 mg/L but showed a toxic effect above 500 mg/L. The (G %) values is 95% for the control (0 mg/L), 95% (10 mg/L), 95 (100 mg/L), 85% (500 mg/L), 85% (1000 mg/L) showed that compound (1) have an influence on seed germination while the (RRG %) ranging from 48.72 to 113.82 showed also negative allelopathic effect on seed germination at high concentrations (≥ 500 mg/L). Iridoids are compounds originally isolated from *Iridomirmex* genus ants in Australia and are involved in the defense mechanism of these ants. Iridoids, are known to have a wide range of biological activities including antibacterial, antispasmodic, and antioxidant properties [31]. We evaluated in this study for the first time the phytotoxicity of an iridoid compound, Shanzhiside methyl ester (1) (Figures: 6-13).

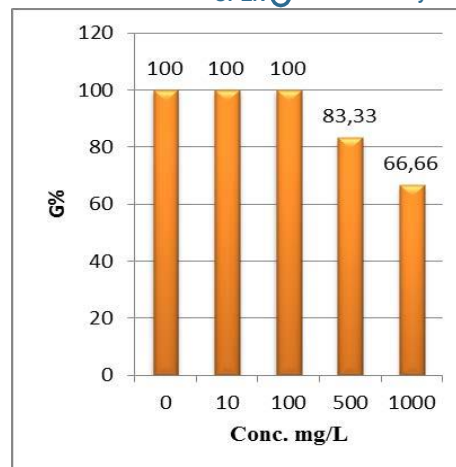


Figure 6: Percentage of germination (G%) of *Lepidium sativum* seeds exposed to different concentrations of root crude extracts.

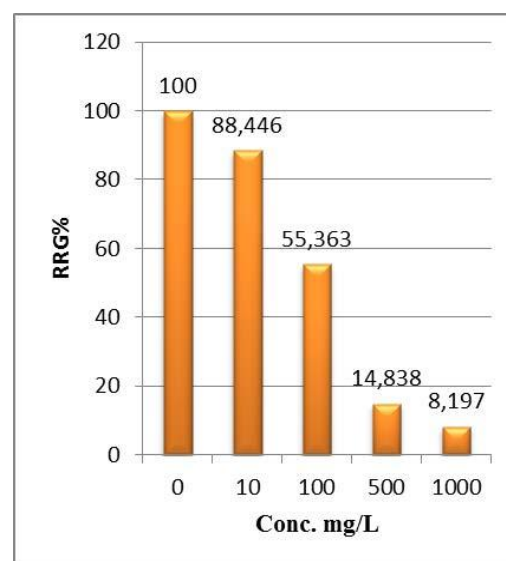


Figure 7: Relative root growth (RRG%) of *Lepidium sativum* exposed to different concentrations of root crude extracts.

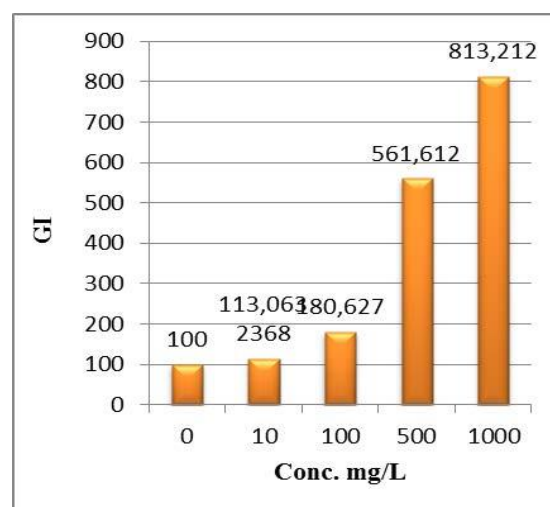


Figure 8: Germination Index (GI %) of *Lepidium sativum* exposed to different concentrations of root crude extracts.

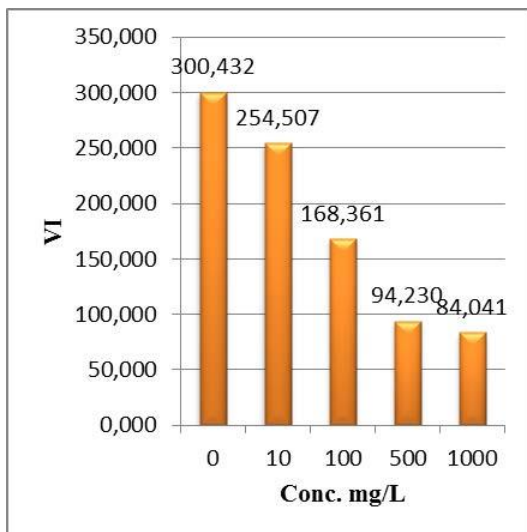


Figure 9: Vigor index (VI) of *Lepidium sativum* exposed to different concentrations of root crude extracts.

Shanzhiside methyl ester (1)

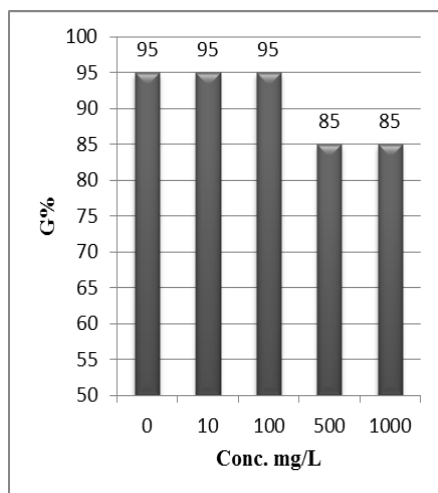


Figure 10: Percentage of germination (G%) of *Lepidium sativum* seeds exposed to different concentrations of compound (1).

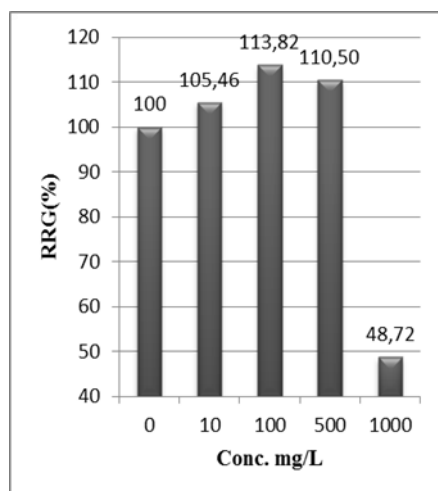


Figure 11: Relative root growth (RRG %) of *Lepidium sativum* exposed to different concentrations of compound (1).

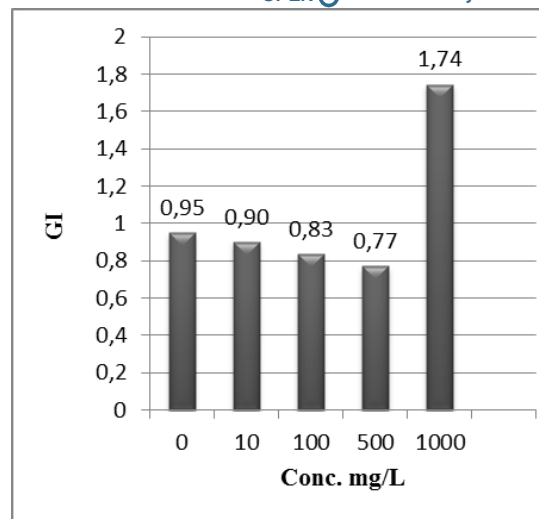


Figure 12: Germination index (GI%) of *Lepidium sativum* exposed to different concentrations of the compound (1).

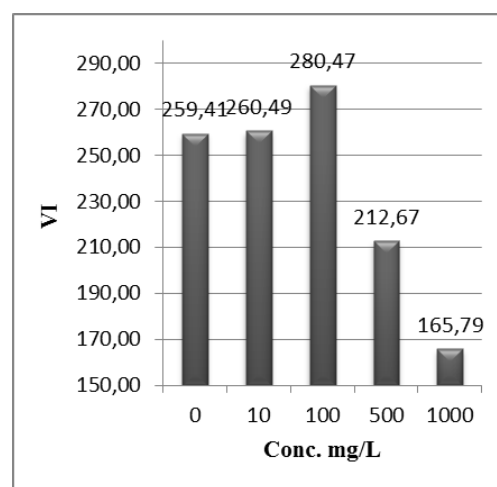


Figure 13: Vigor index (VI %) of *Lepidium sativum* exposed to different concentrations of the compound (1).

CONCLUSION

Phytochemical study of crude extracts from *C. febrifuga* led to the isolation and structural identification of nine compounds belonging to four classes of organic compounds (iridoids, saponins, triterpenes, and sugars). We report the direct isolation of a saponin (3β -(α -L-rhamnopyranosyloxy)-ursa-12,20(30)-diene-27,28-dioic acid) previously obtained by acid hydrolysis of 3β -(α -L-rhamnopyranosyloxy)-28-O-(β -D-glucopyranosyl)-ursa-12,20(30)-diene 27, 28-dioic acid isolated from root extracts of *C. febrifuga*. The tested compound (1), an iridoid, stimulated seedling growth at low doses (10 mg/L-100 mg/L) and exhibited moderate phytotoxicity at 500 mg/L. The stem bark extract was the most effective against the superoxide anion in electrochemical measurements with an AI_{30} of 75.9 mg/L. This study completes the wide range of biological activities of *C. febrifuga* extracts as well as an iridoid compound.

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