

Computational Design of Thermally Activated Delayed Fluorescence Materials: The Challenges Ahead

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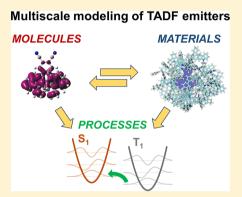
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ABSTRACT: Thermally activated delayed fluorescence (TADF) offers promise for all-organic light-emitting diodes with quantum efficiencies competing with those of transition-metal-based phosphorescent devices. While computational efforts have so far largely focused on gas-phase calculations of singlet and triplet excitation energies, the design of TADF materials requires multiple methodological developments targeting among others a quantitative description of electronic excitation energetics, fully accounting for environmental electrostatics and molecular conformational effects, the accurate assessment of the quantum mechanical interactions that trigger the elementary electronic processes involved in TADF, and a robust picture for the dynamics of these fundamental processes. In this Perspective, we describe some recent progress along those lines and highlight the main challenges ahead for modeling, which we hope will be useful to the whole TADF community.



rganic light-emitting diodes (OLEDs) have emerged as a mature technology, reaching commercial applications in lighting and full-color displays. Delayed emission with characteristic spectra coinciding with prompt fluorescence but differing in the emission lifetimes has been observed in eosin solutions and has since been known as E-type fluorescence.¹ This is also referred to as thermally activated delayed fluorescence (TADF), a process that offers the promise to boost internal quantum efficiency (IQE) of electroluminescence beyond the 25% spin statistical limit (Figure 1). The demonstration of TADF in OLEDs has shifted the material design paradigm from phosphors containing rare and expensive transition metals toward all-organic compounds with reduced singlet-triplet exchange interactions.

TADF emission is triggered by reverse intersystem crossing (RISC) from the nonradiative triplet "reservoir" states to radiative singlet states,² a process that is facilitated by a small energy splitting ΔE_{ST} between the lowest singlet (S₁) and triplet (T_1) excited states, and possibly assisted by the manifold of higher-lying triplet states $(T_n, \text{ see Figure 2})$.³⁻⁵ Hence, both prompt (from the lowest singlet) and delayed (from upconverted triplets) fluorescence decay channels add up in converting into light potentially all electrically generated excitons, irrespective of spin. The most common design strategy of TADF emitters consists in partitioning hole and electron densities over different spatial regions via electron donating (D) and accepting (A) units, often connected in a twisted conformation, hence reducing exchange interactions splitting singlets from triplets.⁶⁻⁹ In actual cases, the excited

states involved in TADF often turn out to be hybrid mixtures of charge transfer (CT) and local excitation (LE) diabatic states, with the amount of mixing prompted by vibronic coupling.^{10,11} We refer the interested reader to a recently published review¹² for a critical discussion of the chemical design rules that have emerged so far through the fruitful interplay between theoretical and experimental chemists. Although some progress has been achieved, molecular design has largely relied on the calculation of excitation energies from the optimized ground-state geometry using time-dependent density functional theory (TD-DFT), either for isolated molecules or assuming a continuum embedding (as for instance described using polarizable continuum models, PCMs).¹³⁻¹⁵ The sole criterion for the selection of potential candidates for TADF is then the calculated exchange gap, ΔE_{ST} , vertical or adiabatic, between the lowest singlet and triplet excitations. Besides raising a number of important technical questions regarding the accuracy of the predictions, this simplified view of the TADF mechanism might considerably bias the engineering rules, as it neglects potentially important effects such as the specifics of intramolecular conformation and intermolecular (host-guest) interactions on spin conversion dynamics in the solid state. This Perspective aims at providing a comprehensive view on the modeling of TADF emitters and at highlighting the challenges ahead. For being predictive, we

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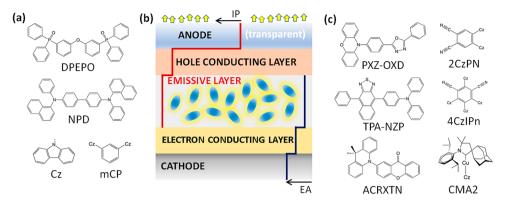
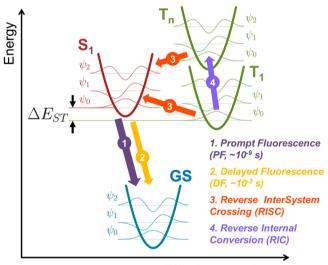


Figure 1. (a) Chemical structures of some relevant host materials: Bis[2-(diphenylphosphino)phenyl] ether oxide (DPEPO), N,N'bis(naphthalen-1-yl)-N,N'-bis(phenyl)benzidine (NPD), and 1,3-di(9H-carbazol-9-yl)benzene, 9,9'-(1,3-Phenylene)bis-9H-carbazole (mCP). Chemical structure of the carbazole (Cz) group. (b) Minimal sketch of a TADF-doped OLED, where electron and hole conducting layers serve also as hole and blocking layers, respectively, and as exciton blocking layers. Dark blue and red lines indicate typical levels for electron affinities and ionization potentials. This Perspective focuses only on the emissive layer, where electrons and holes recombine to form excitons. The layer is constituted of a hole and electron conducting matrix (host), containing a few percent of TADF emitters (guest). The ideal host should possess EA (IP) values lower (higher) than the emitter. (c) Chemical structures of emitters discussed in the text: phenoxazine-2,5-diphenyl-1,3,4-oxadiazole (PXZ-OXD), 4,5-di (9H-carbazol-9-yl) phthalonitrile (2CzPN), N,N-diphenyl-4-(9-phenylnaphtho-[2,3-c][1,2,5]-thiadiazol-4-yl)aniline (TPA-NZP), 2,4,5,6-tetra(9H-carbazol-9-yl)isophthalonitrile (4CzIPN), 3-(9,9-dimethylacridin-10(9H)-yl)-9H-xanthen-9-one (ACRXTN), and cyclic (alkyl)(amino)carbene ligands copper-carbazole (CMA2).



Effective coordinate

Figure 2. Illustration of the electronic states involved in TADF, i.e., ground state (GS) and singlet (S_1) and triplet (T_1, T_n) excited states along an effective vibrational coordinate. The elementary steps leading to prompt (PF) and delayed (DF) fluorescence are illustrated as arrows. Low-quanta vibrational wave functions, which promote reverse intersystem crossing (RISC) and reverse internal conversion (RIC), are also shown.

argue that TADF-oriented computational models should include the following in an integrated framework: $^{10}\,$

- An accurate description of the singlet and triplet excitedstate manifolds in terms of their relative energy and detailed nature, including coupling to relevant vibrational degrees of freedom;
- A reliable evaluation of spin-orbit and hyperfine couplings mediating (reverse) singlet-triplet intersystem crossing;
- (3) A proper embedding scheme to account for both structural and electrostatic effects in the emitting layer;
- (4) A solver for excited-state dynamics that includes electron-phonon and spin-mixing interactions on an equal footing.

In the following, the needed methodologies to reach these ambitious targets are critically reviewed and discussed, going all the way from a quantum-chemical description of the molecular TADF building blocks to the simulation of the corresponding solid-state materials using combined quantumclassical methods.

> Can we predict the nature of the excited states and tune the primary chemical structure of TADF emitters to maximize the efficiency of upconversion and light emission in TADF-based OLEDs?

1. Molecular Electronic Excitations. 1.1. Ab Initio Wave Function-Based Methods. Generally speaking, the set of molecules acting as hosts or emitters share a common feature, that is, a relatively large size compared to those molecules constituting the data sets used for historically benchmarking *ab initio* methods. $^{16-18}$ We focus here on modern variants of ab initio theories for excited states, particularly for calculating vertical transition energies to the lowest excited state of singlet and triplet spin multiplicity, $\Omega(S_1)$ and $\Omega(T_1)$, respectively, also applicable to the whole manifold of S_n and T_n states. Although in principle superior to all other amenable approaches, because of the unfavorable scaling with system size (N) of the family of methods based on iterative coupled-cluster (CC) equations, their application to TADF compounds have been severely limited, and only recently have some examples applying the hierarchy of CC-based methods to small molecules appeared.¹⁹ Actually, for CC-based methods, the hierarchy is CC2 (N^5) < CCSD (N^6) < CC3 (N^7) < CCSDT (N^8) , concerning both computational scaling with system size (N) and expected accuracy. Note also that CC2 (CC3) is an approximation to CCSD (CCSDT) that has been developed with the prediction of excited-state energies as the main focus and that all these

variants are based on linear-response theory. On the other hand, EOM-CCSD (N^6) is based on a slightly different theoretical frame, though it leads to similar excitation energies (yet different transition dipole moments).

The accuracy of the methods is very high, e.g. CCSDT excitation energies are within a few tens of millielectronvolts w.r.t. FCI, but they are amenable to only very small molecules and hence not applicable for most TADF chromophores. However, simplified versions of second-order wave function methods, and thus with a reduced formal scaling of $O(N^5)$ with respect to both canonical equations and higher-order methods such as CCSDT, are available. We mention, among them, linearresponse CC2 and ADC(2) as emerging cost-effective and accurate methods.²⁰ Note that in the case of CC2, because the full expression for double amplitudes is retained only at first order, the method behaves better for single excitations and tends to overestimate excitation energies for doubly excited states. The ADC(2) method suffers from the same inherent limitation for double excitations, and the corresponding ADC(3) is too costly for routine calculations, although it leads to reliable benchmarks. However, the extended ADC(2)x version provides an overall improvement for the description of excited states with double-excitation character. An interesting compromise between accuracy and computational cost is provided by the spin-component-scaling (SCS-)CC2 method through the introduction of scaling factors to the same-spin and opposite-spin contributions of the second-order correlation energy, leading thereby to a better performance for excited states.²¹ There are other methods that allow, by construction, *n*- or multielectron excitations, that is by promoting up to n electrons into an active window of m molecular orbitals, as in RAS- or CASSCF/CASPT2 methods. However, computational limitations preclude the use of large (n, m)active spaces ideally including all relevant π and π^* orbitals. We finally mention many-body Green's function methods such as the GW plus Bethe-Salpeter equation (BSE) formalisms, also explored recently within this context with some success.^{22,23} We shall, however, remark that full BSE suffers the issue of triplet instability, much as TD-DFT does²⁴ (see below).

As other possible caveats for these calculations, we here mention (i) the very costly calculation of excited-state geometries and adiabatic values for $\Omega(S_1)$ and $\Omega(T_1)$, which are important figures when dealing with light emission, and (ii) the marked dependence (i.e., slow convergence) with basis sets size expected for these methods and the possible need to include diffuse functions. This has historically prompted the use of more cost-effective methods, despite the fact that the computational efforts needed for ab initio calculations can be reduced by using resolution-of-the-identity^{25,26} or densityfitting techniques.²⁷ To conclude, to date, wave function-based methods are not routinely used for TADF applications and TD-DFT is usually preferred. However, they offer a robust framework to tackle challenging issues, such as assessing the contribution to RISC of higher-lying singlet and triplet excited states²⁶ or modeling multiple resonance effects.²⁸,

1.2. Density Functional Theory. Because of the need to handle large molecular sizes and to achieve a good trade-off between accuracy, scaling, and feasibility of the calculations for the fast screening of compounds, (time-dependent) density functional theory methods are by far the most employed computational tools in the TADF community. However, contrary to what was initially thought, the reliability of the results depends not only on the functional choice for the time-dependent part but also on the whole computational protocol employed, including geometrical and basis sets issues, as well as on the chemical nature of the target molecule. We focus next on excited-state properties of emitters, for which a relatively wide body of information is available from the recent literature.

Earlier calculations combined the use of the B3LYP functional with a moderate basis set such as 6-31G*, to obtain ground-state geometries and dissect the spatial shape and energy location of the frontier molecular orbitals.^{30,31} When applied to the calculation of vertical transition energies to the lowest excited state of singlet and triplet spin symmetry and their difference dubbed as ΔE_{ST}^V , the results were found to critically depend on the weight, or proportion, of the exact-like exchange introduced into the hybrid functional (typically ranging from 5 to 40%, namely 20% for the widely used B3LYP model).³² Note that any general protocol should give accurate excitation energies for the lowest singlet and triplet excited states without relying on any error cancellation and without limiting the focus to the magnitude of $\Delta E_{\rm ST}^{\rm V}$ as a unique target. Namely, a system- and state-dependent procedure should be avoided, as extension to excited states beyond S_1 and T_1 might be problematic. Because organic molecules are known to benefit from higher-than-defaults (e.g., default values for B3LYP and PBE0 are 20% and 25%, respectively) weights of the exact-like exchange, other functionals like M06-2X have also been used with some success to date.³³

A major breakthrough was found after imposing the Tamm-Dancoff approximation (TDA) for the solution of the full TD-DFT equation,³⁴ which improves the accuracy of routine calculations. This is especially true for T₁ states where TDA cures for the triplet instability problem and, as a result, yields improved ΔE_{ST}^V values. Note that the Thomas–Reiche–Kuhn sum rule for oscillator strengths (i.e., the sum of all oscillator strengths from a particular state to all other equals the number of electrons in the system) is no longer fulfilled with TDA-DFT, as it happened with CIS too. That violation can lead to inaccurate oscillator strength distribution and thus precludes a state-by-state quantitative comparison of oscillator strength values between theories fulfilling (i.e., TD-HF and TD-DFT) or violating the rule. The PBE0 functional was also applied with some correction for dispersion, i.e. D3(BJ), intended mostly to provide more accurate ground- and excited-state geometries after including noncovalent (intramolecular) effects, and used with large basis sets, i.e. def2-TZVP, to estimate $\Omega(S_1)$ and $\Omega(T_1)$ energies at the (nearly) complete basis set limit. However, while accuracy reaching 0.1-0.2 eV can be achieved for ΔE_{ST}^V values,¹³ studies using this computational protocol have not yet been extended beyond S_1 and T_1 states, and thus further efforts are still needed in this direction.

Another strategy is provided by range-separated hybrid functionals, i.e. CAM-B3LYP or ω B97X as paradigmatic examples, in which the range-separation parameter may be fine-tuned for each compound,^{14,35} as well as for isolated or host-embedded emitters. In these range-separation models, the electron–electron interaction is split into two contributions, short- and long-range, treating each one at a different theoretical level (short-range often with a semilocal GGA exchange functional and long-range with exact-like exchange to obtain the desired correct asymptotic behavior). The first mention in the literature was by Savin et al. in 1997,³⁶ but for the coupling of multiconfigurational *ab initio* with DFT methods, with probably the first application to GGA exchange functionals by

Table 1. Summary of the	Metrics Most Used and	Their Limiting Values in th	he Case of Pure CT and LE Excitations

		distinction between the nature of excited states	
metric	description	СТ	LE
Λ_{ia}	overlap between the <i>ia</i> pair of the norms of molecular orbitals	$\Lambda pprox 0$	$\Lambda \approx 1$
Δr	coefficient-weighted hole-electron distance between a set of orbital centroids	$\Delta r > 2 \text{ Å}$	$\Delta r < 2$ Å
$D_{\rm CT}$	distance between barycenters of density variations (average between the corresponding barycenters)	t – $D_{\rm CT} >$ 1.6 Å	t – $D_{\rm CT}$ < 1.6 Å
$\phi_{ m S}$	normalized overlap between attachment/detachment densities	$\phi_{\rm S} pprox 0$	$\phi_{\rm S} pprox 1$

Hirao et al. in 2001^{37} and popularized later by Handy et al. with the CAM-B3LYP method.³⁸ Because the tuning of the ω parameter aims at accurately reproducing one-electron attachment/detachment energies, it also naturally leads to accurate HOMO–LUMO gaps and corresponding excitedstate energies (relying heavily on that energy difference, to first order). However, more work is still needed to confirm the accuracy of these methods for higher-lying singlets and triplets.

S1 and T1 optimized excited-state geometries are now routinely computed from linear-response TD(A)-DFT. However, different options such as the use of UKS (for T_1) or ROKS (for S₁ and T₁ states) exist.³⁹ Still, we stress that mixing different levels of theory such as TD(A)-DFT (for S1) and UKS (for T_1) can lead, in some instances, to spurious negative values of $\Delta E_{\rm ST}$.^{14,40} The fast screening of TADF molecules would also benefit of low-cost TD-DFT-based methods, such as for example sTDA-DFT or sTDA-xTB.⁴¹ It would also be desirable to extend these methods beyond one-electron effects. The use of a (D)-like correction introduced by double-hybrid methods, namely a MP2-like term added non-self-consistently to the standard TD-DFT treatment or the coupling of DFT with MRCI (DFT-MRCI), are among the envisioned possibilities, although the price to pay is a higher formal scaling, $O(N^{\circ})$ in the case of double-hybrid functionals. While doublehybrid models appear robust for $\Omega(S1)$ values, the lack of implementations for the calculation of the corresponding $\Omega(T1)$ values has precluded so far the evaluation of ΔE_{ST}^{V} . DFT-MRCI has been recently extended to deal with higherorder excitations in multichromophoric systems,⁴³ opening new possibilities for exploring light-emission mechanisms for which both singlet and triplet states are of importance, as well as their possible modulation by environmental effects.⁴

1.3. Metrics for the Nature of Electronic Excitations. The need for quantifying the nature of the excited states involved in the TADF process has led to the development of a set of metrics able to gauge the CT versus LE character of the excitations, which is critically entangled with the molecular geometry and its fluctuations around equilibrium.^{10,43,45} The existing metrics can be classified into two groups: those based on molecular orbitals, be they KS or NTO, and those based on density differences. Note that the metrics are intended to be (i) generally applicable, in the sense that they can be applied to any of the desired singlet or triplet excited states, and (ii) easily transferable, with the results not expected to heavily depend on the functional and/or basis set choice. However, the use of relaxed and unrelaxed density might introduce some difference from a quantitative standpoint,⁴⁶ as was recently disclosed.

Table 1 summarizes the observables developed so far and their typical values for the limiting cases of pure CT and LE. As regards the first category of tools, the metrics Λ roughly measures the overlap between a pair of occupied (*i*) and virtual (*a*) orbitals involved in the ground-to-excited state transition;^{47,48} other approaches, such as Δr ,^{49,50} aim at giving an effective

hole–electron separation during the excitation. On the other hand, for density-based descriptors, the $D_{\rm CT}$ index is based on the barycenters of densities associated with an electronic transition,⁵¹ while $\phi_{\rm S}$ refers to the overlap between the attachment/detachment densities, that is, the electron density removed or rearranged during the excitation.⁵² In some cases, because of a spurious behavior of density functionals for CT states, low-lying but unreal (intruder) CT states can appear, which can be also identified and discarded on the basis of the evaluation of intermolecular electrostatic interactions, e.g. by the $M_{\rm AC}$ metrics.⁵³ Despite the appearance of a few topological metrics in the last years, there is also convincing evidence that these are highly correlated, as recently shown for a large sample of molecules displaying CT excitations,⁵⁴ which should facilitate comparison between different existing or future studies.

Metrics based exclusively on orbitals raise some concern. For instance, many-electron excited states might involve multiple pairs of orbitals, thus casting doubt on the meaning of the Λ value. Another example is the case of a symmetric D–A–D triad, where the hole and electron centroids can spuriously occupy similar spatial positions, hence leading to a vanishing value of Δr , despite a strong CT excited-state character. $D_{\rm CT}$ and $\phi_{\rm S}$ overcome these limitations and appear as more universal metrics.⁵⁵

Interestingly, the use of these metrics has clearly evidenced that the excited states involved in the TADF process are neither full CT or LE, as improperly claimed when discussing the photophysics of TADF emitters,^{3,5} but most often feature a rather mixed CT–LE character^{7,10,56} (see Figure 3a). The amount of CT–LE mixing is controlled by the magnitude of the D–A electronic coupling, which is in turn governed by structural parameters (such as the torsion angle between the two moieties).^{7,10} While the S₁ excited state often displays a large CT character (small ϕ_S^7 and large Δr^{13} , see Figure 3a), especially in conformations with near orthogonal D and A moieties,⁶ this is less true for the T₁ excited state. This arises from exchange interactions that stabilize localized triplets more than their singlet counterparts, thereby prompting a more intimate LE–CT mixing in the triplet manifold.^{7,57}

From a practical point of view, the metrics presented in Table 1 have been applied as a tool to rationalize the ΔE_{ST} and oscillator strength values in TADF compounds. Especially, it has been evidenced that the nature of T₁ is the limiting factor in order to minimize ΔE_{ST} . The nature of T₁ can be directly probed experimentally by extracting the zero-field splitting (ZFS) parameters from electron spin resonance (ESR) spectroscopy studies. The dominant dipole–dipole component to the ZFS parameters is indeed inversely proportional to the third power of the interspin distance.⁵⁸ The larger the effective electron–hole radius and spin–spin distance and the lower the ZFS parameter value. From a computational point of view, the ZFS parameter can be evaluated through unrestricted

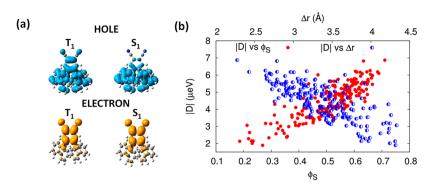


Figure 3. (a) Hole and electron densities calculated in the attachment/detachment formalism for T_1 and S_1 excited states for 2CzPN. (b) Absolute value of the T_1 zero-field splitting parameter as a function of the overlap between the hole and electron densities, ϕ_{s} , and the distance between the hole and electron densities centroids, Δr , as calculated in the attachment/detachment formalism for 2CzPN T_1 excited state. Reprinted and adapted with permission from ref 10. Copyright 2017 American Physical Society.

DFT calculations using EPR-II and EP-III basis sets that are optimized for the computation of hyperfine coupling constants.⁵⁹ Interestingly, a recent study of carbazolyl-dicyanobenzene-based TADF emitters (2CzPN and 4CzIPN, see chemical structures in Figure 1) has demonstrated that ZFS (as calculated at the UKS level) and $\phi_{\rm S}$ (obtained from TDA-DFT) values go together, offering the possibility of comparing experimental and theoretical metrics of the triplet excitations⁶⁰ (see Figure 3b). As for singlet excitations, calculated oscillator strengths have been shown to correlate very well to Δr^{11} and overlap metrics such as Λ_{ia}^{61} and $\phi_{\rm S}^{.7}$

Finally, we mention that only some of the metrics above can be calculated with common quantum chemistry codes, while others require postprocessing after the TD(A)-DFT run. More automated procedures and further benchmarking are definitively needed, as well as their integration in large-scale computational codes.

2. Nonradiative Processes. Nonradiative processes are playing a key role in determining the internal quantum efficiency of TADF-based devices. On the one hand, reverse intersystem crossing (RISC) and reverse internal conversion (RIC) are promoting the upconversion from the triplet to the singlet excited-state manifold. On the other hand, ISC favors the generation of triplets that could subsequently be recycled to S_1 or, because of their long lifetime, could decay through triplettriplet or triplet-polaron annihilation processes, giving rise to device efficiency roll-off at high luminance. Of course, also nonradiative decays to the ground state from both the lowest singlet and triplet states are competing pathways reducing IQE. In the following, we distinguish between spin-conserving, namely (R)IC, triplet-triplet and triplet-polaron annihilation, and spin-nonconserving processes, namely ISC and RISC. Conformational effects on light emission and ISC and RISC are discussed thoroughly below.

2.1. Singlet-Triplet Interconversion. Spin-orbit is a relativistic effect that is responsible for mixing orbital and spin degrees of freedom, thus allowing electronic states of different multiplicities to couple. Spin-orbit coupling naturally arises from the one-electron Dirac equation. Even though being of a fundamental interest, it cannot be solved exactly when considering large many-electron systems such as TADF molecules. Practical applications thus rely on perturbative approaches to nonrelativistic electronic structure calculations. Among the most commonly reported approaches, we distinguish the zero-order relativistic approximation (ZORA) from the full Breit-Pauli Hamiltonian (including relativistic mass corrections and

spin-orbit effects) and its mean-field approximation that are implemented in a number of software packages based on CASSCF,⁶² TD-DFT, and DFT-MRCI⁴³ excited-state descriptions. The interest in spin-orbit coupling calculations in the field of OLEDs dates from the emergence of phosphorescent emitters containing heavy metal centers allowing for rapid ISC from S_1 to T_1 and radiative decay from T_1 to the ground state (usually in the microsecond time scale regime). Because of the presence of these heavy atoms, spin-orbit coupling is usually large in phosphorescent emitters allowing for strong mixing between singlet and triplet excited states, while it is relatively weak (on the order of tenths or hundredths of millielectronvolts) for TADF emitters made only of light elements. Still, the RISC mechanism is expected to be mainly driven by spinorbit coupling with a small contribution of hyperfine coupling, as evidenced by electron paramagnetic resonance measurements.⁹ This is further supported by transient electron spin resonance (TrESR) on 2CzPN and 4CzIPN carbazolyldicyanobenzene-based TADF emitters, for which the observed absorption and emission patterns are characteristic of a spinorbit-driven ISC mechanism. To date, a systematic comparison between the different software and models has not been carried out yet, although it was shown that the full Breit-Pauli model and its mean field approximation lead to very similar spinorbit coupling matrix elements for organic dyes.⁶³ Already in the 1960s, El-Sayed⁶⁴ highlighted the fact that spin-orbit coupling is vanishing when (R)ISC takes place between excited states with identical $\pi - \pi^*$ character. This is because the change in spin angular momentum must be compensated by a corresponding change in angular orbital momentum so that total angular momentum is conserved. As a result, only excited states with different spatial wave functions couple through spin-orbit coupling. A textbook example is benzophenone that sustains $n-\pi^*$ and $\pi-\pi^*$ electronic excitations coupled through large spin-orbit matrix elements.⁶⁵ Recently, this concept was put in practice in the context of TADF materials by associating the magnitude of the spin-orbit coupling to the difference in the nature of these excited states calculated based on the difference of normalized overlap between attachment/detachment densities, $\Delta \phi_{S}$, between S₁ and T₁. $\Delta \phi_{S}$ was calculated on molecular conformations taken from an amorphous morphology of pure films of 2CzPN and 4CzIPN. Spin-orbit matrix elements are found to correlate approximately linearly with $\Delta \phi_{\rm S}$, becoming vanishingly small for $\Delta \phi_{\rm S} = 0$, namely in the case where S₁ and T₁ are nearly degenerate ($\Delta E_{ST} \sim 0$) and exhibit both a strong and identical CT character, in line

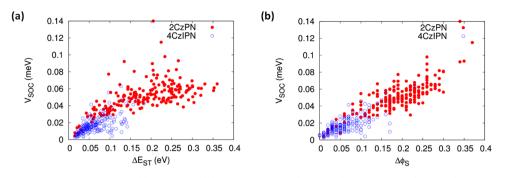


Figure 4. Spin–orbit coupling as a function of (a) ΔE_{ST} and (b) $\Delta \phi_S$ in 2CzPN (blue data) and 4CzIPN (red data), as sampled from a realistic amorphous morphology simulated with molecular dynamics. The approximately linear correlation between V_{SOC} and $\Delta \phi_S$ follows from the El-Sayed rules, while the relationship between V_{SOC} and ΔE_{ST} highlights the need for a trade-off between different parameters for the optimization of TADF performances. Reprinted with permission from ref 10. Copyright 2017 American Physical Society.

with the El-Sayed rules. Interestingly, because of its overall larger $\Delta \phi_{\rm S}$ offset, spin—orbit coupling is larger for 2CzPN than for 4CzIPN (see Figure 4b). Most importantly, we note that in the case of the carbazole derivatives studied in ref 7, spin—orbit coupling and $\Delta E_{\rm ST}$ have antagonistic evolution; such a trade-off has to be found in order to maximize the rate of RISC (see Figure 4a).

2.2. Spin-Conserving Nonradiative Processes. 2.2.1. (Reverse) Internal Conversion. Two different internal conversion mechanisms with contrasting effects are considered here. On the one hand, the reverse IC in the triplet manifold of excited states, which is a thermally activated process, is expected to assist RISC by promoting the formation of higher-lying triplet excited states, from which the conversion to the singlet manifold might occur more efficiently because it is associated with a larger exergonic character. This interconversion channel is expected to compete with the direct conversion from T_1 to S₁ and features smaller activation energies.⁶⁶ On the other hand, nonradiative recombination to the ground state should take place mainly from S1 and contribute as the main monomolecular pathway to molecular excitation loss. Even though the two IC processes described above involve excited states of different spin multiplicities, the initial and final states are in both cases coupled through nonadiabatic couplings. The rate of internal conversion has been calculated in triphenylaminethiadiazole molecule (see chemical structure in Figure 1) using a Fermi golden rule formalism and considering nonadiabatic couplings for all relevant vibrational normal modes.⁶⁷ This approach, however, breaks down in the case internal conversion occurs at conical intersections, with points of degeneracy between electronic states acting as dynamic funnels for radiationless transitions.68,69

How important is the simulation procedure in determining the final morphology, and in turn how does it influence the calculated electronic properties?

2.2.2. Triplet-Triplet and Triplet-Polaron Annihilation. In an OLED device, at high current, i.e. high hole and electron densities, triplets start accumulating in the device. Bimolecular processes such as triplet-triplet or triplet-polaron annihilation might occur, leading to an undesired roll-off behavior.⁷⁰ To date, no mechanistic picture has been proposed for these

processes in the context of TADF materials. Triplet-triplet annihilation can proceed either via a virtual CT state or through a two-electron exchange mechanism, similarly to its reverse process, singlet fission.⁷¹ The first attempts to include both processes in a device-like kinetic Monte Carlo simulation have been based on phenomenological grounds considering that triplet-triplet (triplet-polaron) annihilation takes place when two triplet excitations (a triplet excitation and a charge) occupy neighboring sites.⁷² Assessing annihilation rates from first principles remains very challenging because these processes involve the transient formation of high-lying electronic excitations with multiple-excitation character, the description of which demands highly correlated quantumchemical methods often difficult to handle for large-size TADF emitters. A further challenge is that because annihilation is a bimolecular process, its theoretical investigation requires the knowledge of the microscopic arrangement of emitters and thus a further layer of calculations, as detailed in the next paragraphs.

3. From Molecules to Materials. 3.1. Morphology of OLED Active Layers. With respect to the gas phase or implicit solvent calculations described so far, a step further toward the realistic modeling of TADF materials consists in taking into account explicitly the presence of surrounding host molecules. This is possible via a simple two-step multiscale scheme, by first employing classical (i.e., molecular mechanics) force fields and molecular dynamics (MD) or Monte Carlo (MC) simulations to produce a realistic guess of the emitting layer morphology (see Figure 1 for a simplified scheme of a multilayer OLED architecture)⁷³ and then performing electronic structure calculations on single molecules or clusters extracted from the simulated trajectories. This approach is considerably more expensive than a pure quantum mechanical (QM)based study, but its higher computational cost is compensated by the number of additional important effects occurring in real devices that can be accounted for, namely, (i) sampling of many molecular geometries with a probability of occurrence depending on temperature (Boltzmann-weighted), (ii) conformational changes and freezing induced by the solid matrix, and (iii) inclusion of polarization and electrostatic effects. In addition, simulations predict the distribution of emitter orientations and, more generally provide the molecular positions of carriers and emitters, which are fundamental ingredients for device modeling, and in particular for kinetic simulations of electronic processes.

Note that emitting layers are typically (meta)stable, disordered or partially ordered glasses, and host and emitter

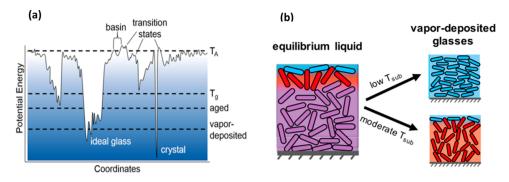


Figure 5. (a) Schematic representation of the potential energy landscape of a glass-forming system. T_A is the temperature where non-Arrhenius dynamics are first observed in the liquid. Upon cooling at a constant rate, a temperature is reached at which molecular motions freeze and a glass is formed (glass transition temperature, T_g). Aging slightly below T_g allows some equilibration and lowers the potential energy. For some systems, much lower energies can be reached by physical vapor deposition with respect to aging or slow cooling. (b) Possible origin of anisotropic molecular packing in vapor-deposited glasses of a rod-like molecule for which the free surface of the equilibrium liquid is anisotropic.^{78,88} The substrate temperature T_{sub} determines the depth to which structure at the surface can equilibrate during deposition. The lowest portion of the growing film becomes trapped by further deposition. Also the interface with the solid substrate can be in principle anisotropic. Reprinted with permission from ref 83. Copyright 2017 AIP Publishing.

molecules themselves (Figure 1) have characteristics of organic glass-formers, being often constituted of a few nonlinear rigid aromatic units interconnected by single rotatable bonds, allowing thus for a relatively large number of stable conformations. There are at least three routes to simulate amorphous morphologies of emissive layers, all of them essentially out of thermodynamic equilibrium. The simplest scheme starts with an NPT-ensemble MD simulation at temperatures and pressures high enough to obtain a fluid mixture of the desired composition and subsequently cools the sample at room temperature and equilibrates it until average volume and total energy appear stable in time.^{10,75} A second, more costly approach imitates a solvent annealing process by starting from a concentrated solution of the target materials equilibrated at ambient conditions, from which molecules of solvent are progressively removed until a dry film is obtained. More recently, MD or MC simulations in which molecules are progressively inserted in the system and landed on a substrate, a scheme pioneered by some of us for the vapor deposition of organic crystalline semiconductors,^{76,77} have gained increasing popularity in the OLED research field.^{78–81} The origin of this trend is twofold: on the one hand, real TADF-based active layers are in fact prepared most often by codeposition of (at least) one host semiconductor and a guest TADF emitter,⁸² and on the other hand, because these films are amorphous, it is difficult to validate the simulation results versus experimental structural data, and it is tempting to believe that mimicking the experimental process could improve the quality of the predictions. Indeed, one of the open questions in the field is how important is the simulation procedure in determining the final morphology, and in turn how does it influence the calculated electronic properties. Experimentally, there is a large body of evidence suggesting that vapor-deposited systems are in general more stable and dense than spin-coated ones prepared by solvent

> How can we design in silico TADF emitters that yield the right orientation in the solid-state matrix to maximize light outcoupling?

evaporation⁸² or freezing from the liquid phase;⁸³ however, the exact shape of the potential energy landscape (see scheme in Figure 5a for a pure material) depends not only on the chemical nature of the host-guest system but also on their relative concentration.

A further important result has recently emerged from both experiments^{78,84,85} and simulations:⁷⁸⁻⁸¹ vapor deposited-glasses can be to some extent orientationally anisotropic or, in other words, the orientation of the emitters in the active layer is not completely random. Because the specific orientation of emitter transition dipoles can strongly impact the outcoupling efficiency,^{86,87} it would be desirable to employ computer simulations in designing materials and setups apt at precisely controlling the orientation of the dyes, and then the direction of emitted light, in order to maximize the light output. The insurgence of anisotropy has been attributed to multiple factors, ranging from van der Waals interactions to dipole-dipole interactions and kinetic effects. These may be again system-dependent, but regardless of specific effects, it appears clear that a critical role is played by the orientation of the molecules at the growing interface with vacuum, as depicted in Figure 5b. Unfortunately, the current lack of knowledge about the relationship between all the simulation parameters (e.g., models, rates, temperatures, times, etc.) and the morphology obtained, so far does not allow establishing how general are the above-mentioned effects and to which extent they could be applicable to experimental results. Therefore, more systematic studies in this direction are urgently needed.

Another, maybe unavoidable, problem in comparing simulations and experiments is the enormous gap between the time scales accessible to simulations (microseconds) and the ones employed in the lab (minutes or hours). This difference might shift the balance between kinetic and thermodynamic effects and may hamper the successful application of computer simulations to systems in which this balance is delicate. Concerning instead length scales, the mismatch between reality and atomistic models is much narrower, hundreds of nanometers versus tens or so. To reach the device scale, coarse-grained models can be employed,^{78,89} with the main disadvantage of some extra effort for the parametrization and complicated backmapping schemes to revert to the fully atomistic morphology, needed for electronic structure calculations. Unitedatom force fields can be an effective compromise, because the

reduced number of centers allows a CPU time speed-up of about 1 order of magnitude⁹⁰ and the backmapping to a full atom model is straightforward. Focusing on full-atom models, there is a raising awareness that the use of classical force field geometries for QM calculations may lead to uncontrolled approximations and systematic errors in the evaluation of electronic properties, e.g. S₀ and T₁ energies.^{10,75} In order to minimize this source of error, it would be beneficial to employ, in future studies, nontransferable force fields specifically tailored for reproducing not only the QM optimized geometry but also the vibrational frequencies in the electronic state of interest.⁹¹

How much is torsional dynamics affected by the host-guest interaction? How does it affect the efficiency of TADF?

3.2. Conformational Effects. Radiative emission cross sections, also quantified through the oscillator strength, are often very small in TADF molecules with close to orthogonal donor and acceptor moieties in their ground- or excited-state optimized geometries (see Figure 6b). Unless counterbalanced by specific intramolecular interactions such as hydrogen bonds,⁶¹ most TADF molecules usually feature a twisted equilibrium conformation, which inevitably leads to a small overlap between the frontier π -conjugated orbitals of the donor and acceptor moieties. Fortunately, in TADF emitters based on D and A moieties connected through a single bond, soft torsional

modes are easily activated, generating a large spectrum of conformations at room temperature^{4,7} (see Figure 6a). These display significantly different absorption/emission energies and associated $\Delta E_{\rm ST}$ values as the nature of the lowest singlet and triplet excited states (as probed through $\phi_{\rm S}$) varies in time and space (see Figure 6c,d). Thermal excitation of these soft vibrational modes also results in broad and unstructured absorption and emission spectra. Thermal motion around the twisted equilibrium structures has a positive impact on luminescence, as it allows the sampling of conformations with larger overlap between the electron and hole wave functions, i.e. also larger oscillator strengths and radiative decay rates.²⁰ In this case, emission of the TADF materials appears as a vibrationally assisted process strongly coupled to soft (low-frequency) torsional modes.⁷

Also the RISC process, as claimed by Monkman and coworkers,³ can be seen as a spin-vibronic mechanism where spin mixing is dynamically gated by conformational fluctuations triggered by low-energy torsional modes. The influence of torsional modes on excited-state dynamics is twofold: (i) It brings excited states of the same spin multiplicity closer to each other, enhancing nonadiabatic coupling between the lowest singlet (triplet) excited states.⁹² (ii) It affects dynamically the nature of the singlet and triplet excited states involved in the TADF process in such a way that the thermally averaged spin– orbit coupling is enhanced compared to its value at the equilibrium geometry.^{67,92} The role of vibrations on TADF dynamics can be modeled using different formalisms, either based on nonadiabatic molecular dynamics, such as mixed quantum–classical surface hopping methods⁹³ or full quantum

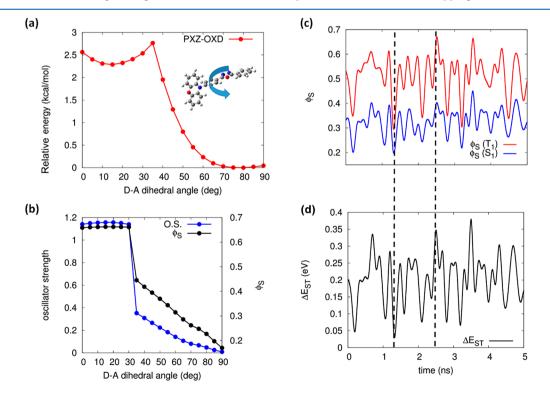


Figure 6. (a) Torsional energy profiles calculated at the PBE0-D3(BJ)/6-31G(d,p) level of theory with the PCM module for solvent (toluene) for PXZ-OXD (see chemical structure in Figure 1). (b) Variation of ΔE_{ST} and oscillator strength (OS) as a function of the D–A torsion angle. (c) Time evolution of the $\phi_S(S_1)$ and $\phi_S(T_1)$ from electronic structure calculations performed along a molecular dynamics trajectory for 2CzPN. (d) Time evolution of the ΔE_{ST} for 2CzPN. Vertical dashed lines highlight that ΔE_{ST} is the largest (smallest) when the difference in $\phi_S(S_1)$ and $\phi_S(T_1)$ is the largest (the smallest). Panels a and b reproduced with permission from ref 7. Copyright 2017 The Royal Society of Chemistry. Panels c and d reprinted and adapted with permission from ref 10. Copyright 2017 American Physical Society.

wavepacket propagation simulations,⁹⁴ or relying on rate expressions derived from time-dependent second-order perturbation theory (Fermi golden rule).^{67,92} These approaches usually include a set of preselected intramolecular vibrational modes for which the frequencies and displacements (electronphonon couplings) are computed from first principles. Alternatively, one can resort to MD simulations that allow sampling all vibrational modes classically, being intra- or intermolecular in origin, at once. In the case of low-frequency vibrations, classical and quantum approximations yield similar results for thermally averaged spin-orbit coupling and $\Delta E_{\rm ST}$ values.⁹⁵ The reorganization energy associated with high-frequency vibrations, i.e., mostly bond stretching, can be obtained from ground- and excited-state geometry optimization of the isolated molecules. These modes steer temperature-independent quantum tunneling effects and can be easily incorporated into rate expressions for ISC and RISC. For instance, by a careful analysis of the time evolution of the dihedral angles in 2CzPN and 4CzIPN, a characteristic time scale of about 1 ps has been inferred. This is fast compared to RISC; hence, a large portion of the dihedral angles distribution is explored by the molecule before upconversion takes place, which confirms the truly dynamic nature of the RISC process. Considering thermally averaged spin-orbit coupling and ΔE_{ST} for the specific case of 4CzIPN, (R)ISC rates calculated within the semiclassical Marcus theory were found to be in excellent agreement with experimental data.¹⁰

To close this discussion, we briefly refer to studies pointing to the role of "hard" modes. For instance, it has been shown that the displacement along a C=O stretching mode on the donor moiety of a xanthone-acridine D-A complex (see chemical structure in Figure 1) is able to bring in near resonance triplet ³CT and ³LE states, from which efficient RISC to ¹CT proceeds.⁹⁶ Similarly, highly correlated wave function-based calculations in carbene-metal-amide complexes (see chemical structure in Figure 1) suggest that the dynamic red shift observed experimentally is associated with changes in the carbon-nitrogen bond length and metal-carbon-nitrogen bond angle within the carbene-metal-amide three-center core. Very interestingly, these changes reduce ΔE_{ST} while keeping unaffected the spin-orbit coupling and emission transition dipole moment from the singlet excited state, at odds with the initially proposed rotationally assisted upconversion mechanism.45

3.3. Electronic Polarization Effects. Environmental effects are known to have a major impact on molecular excitations of charge-transfer character, yet the implications on ΔE_{ST} in TADF emitters have started to be appreciated only recently. The role of the environment is particularly subtle and important in systems where singlet and triplet excitations of CT and/or LE character are all close in energy and compete, while the energy window interesting for applications is a few tens of millielectronvolts. Viable computational strategies consist in QM/MM approaches, where the TADF emitter (QM system) is embedded in a classical medium that can be described either with PCM or with discrete schemes of

> Polarization medium effects stabilize CT states with respect to LE states. Could we possibly target the right combination of host and guest to get negative ΔE_{ST} ?

atomistic resolutions. Both approaches have advantages and limitations that we briefly discuss in the following.

In this context, Brédas and co-workers systematically employed the PCM in their TD-DFT investigations of singlet-triplet splitting and spin-orbit matrix elements for different TADF molecules.^{14,56} Sun et al. specifically addressed the effect of the dielectric constant (ε) on the nature of the excited states and on the single-triplet splitting by using optimally tuned rangeseparated hybrid functionals, whose range separation parameter ω was optimized for each value of ε .⁹⁷ These calculations highlighted the role of the medium polarizability in stabilizing electronic configurations with large CT character, in turn affecting the CT-LE hybridization in singlet and triplet excitations. For instance, dipolar D-A molecules such as TXO-TPA and TXO-PhCz, which in the gas phase are prescribed to feature large ΔE_{ST} (ca. 0.5 eV) as a result of the large LE character of T_1 , become interesting for TADF applications in a typical organic matrix with $\varepsilon \approx 3$. Indeed the medium polarization leads to excitations with large CT character and nearly nonoverlapping hole and electron clouds and $\Delta E_{\rm ST}$ below 0.1 eV.⁹⁷ Marian and Föller⁴⁴ proposed a reinterpretation of the emission observed for carbene-metalamide in both solution and film based on a combination of DFT-MRCI and PCM. Especially, they showed that solvent reorganization has to be taken into account when computing the delayed fluorescence spectrum, while it is not required when discussing prompt fluorescence occurring at an early time scale. Practically speaking, one needs to consider the relaxed density matrix of the excited state to calculate the solvent reaction field. Delayed fluorescence appears to be redshifted in comparison to the prompt one in agreement with experiment. However, in glassy films, such a treatment is not needed because the medium reorientation is sterically hindered so that delayed fluorescence appears to be blue-shifted compared to solution. Along the same line, Penfold et al.⁷² have highlighted through a combination of MD and TD-DFT calculations that a blue shift in the delayed emission of D-A TADF emitters in films at the longest time scale does not result from host reorganization, and thus on specific host-guest interaction, but rather from a distribution of CT states with different emission energies. The prompt fluorescence is essentially governed by the higher-energy CT states that exhibit the largest hole-electron wave function overlap and therefore also oscillator strength. As for the delayed fluorescence, the early part of the signal appears to be red-shifted in comparison to the prompt fluorescence because RISC occurs first through the lower-energy CT states that minimize ΔE_{ST} . The late delayed fluorescence component then occurs through higher-lying but more emissive CT states, thereby rationalizing the blue-shift observed.

Beyond PCM, atomistic polarizable models aim at describing excitations of molecules in their specific environment, usually an amorphous matrix blending the charge transporting material and other emitters. Atomistic simulations of such blends are a prerequisite for these approaches, which also allow for the sampling over a statistical collection of molecule–environment configurations. Our group adopted this route to study environmental effects in carbazolyl-dicyanobenzene-based TADF emitters, 2CzPN and 4CzIPN, employing atomistic microelectrostatic models parametrized from first principles.¹⁰ These calculations showed that the medium polarization, modeled with mutually interacting anisotropic polarizabilities, reacts differently to states of different CT character (see Figure 7a).

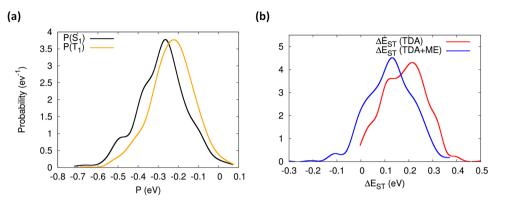


Figure 7. (a) Polarization energy distributions associated with S_1 and T_1 excited states. (b) Distributions of ΔE_{ST} . Red and blue lines correspond to TDA-PBE0 results in the vacuum and accounting for local dielectric effects, respectively. All data provided in this figure refer to 2CzPN. Reprinted and adapted with permission from ref 10. Copyright 2017 American Physical Society.

For instance, 2CzPN shows a singlet excitation S_1 that has larger CT character than the triplet T₁ and thus a larger electrical dipole, resulting in a stronger stabilization of the former by the environment polarization, finally leading to a reduction in ΔE_{ST} (see Figure 7b). This provides a general mechanism through which the embedding medium compensates for large ΔE_{ST} established at the molecular level, leading in some cases to negative ΔE_{ST} values. In addition, atomistic models do also account for the fact that molecules are distorted in real morphologies and experience the inhomogeneous electrostatic potential of the neighborhood. These phenomena affect the energies of the excited states within both the singlet and the triplet manifold of states, leading to inhomogeneous broadening in the solid matrix and to broad distributions of ΔE_{ST} values. Such a disorder can also break the symmetry of the molecule (e.g., localizing the hole of a CT excitation on a given D carbazole unit) and, most interestingly, have a dynamic nature, i.e. lead to a modulation in time of the nature of the relevant states (CT–LE hybridization) and of the $\Delta E_{\rm ST}$.¹⁰

4. Concluding Remarks. In summary, predictive modeling of TADF emitters requires an integrated multiscale approach able to capture the energetics and the dynamics of electronic excitations in realistic morphologies of OLED emitting layers. In combination with state-of-the-art experimental investigations (optical spectroscopy and device characterization), computational studies have already shed light on some key features of TADF, including the following:

- The excited states involved in TADF often feature mixed CT and LE character.
- The host influences $\Delta E_{\rm ST}$ through conformational and dielectric effects.
- RISC and light emission are dynamic processes assisted by intramolecular vibrations.

While some of the items above remain open to discussion, as they are likely material-specific, they also prompt new questions and challenges that need to be addressed when designing the next generation of TADF emitters. For instance, can one tune the nature of the excited states involved in the TADF process in order to speed up RISC? Can one take advantage of environmental effects to design molecules with negative $\Delta E_{\rm ST}$? Can we expand the modeling approaches in order to account for all the mono- and bimolecular radiative and nonradiative processes taking place in TADF-based OLEDs and identify host–guest combinations that would minimize annihilation and maximize pure and color-tunable light emission?

Should we freeze out the motion of triplet excitations in TADF materials to reduce annihilation processes?

Clearly the answer to these questions can be obtained only in the scope of a multifaceted theoretical framework, where molecular and material properties are conjointly addressed and optimized. More specifically, we end this Perspective with modeling challenges inspired by a few selected opportunities from recent experimental investigations:

*Hyperfluorescent OLEDs.*⁹⁸ Here, TADF molecules act mostly as assistant dopants that drive the excitations toward a dye with narrow-line singlet emission. To further improve what is referred to as the fourth-generation OLEDs, a microscopic picture for the diffusion of singlet and triplet excitations relevant to TADF is definitively needed. This would require going beyond the widely used Förster model for weakly dipole-allowed CT singlets and including both exchange and super-exchange interactions for triplets.^{99–101}

Highly Emissive TADF Emitters. Architectures sustaining multiresonance effects^{28,102,103} have the potential to solve the conundrum of large singlet radiative decay rates despite small exchange energies. A proper description of the singlet and triplet excitations in these molecules calls for the inclusion of high-order electronic correlation effects, which are difficult to capture using conventional TD-DFT methods.

Exciplexes. Despite considerable efforts to establish them as viable technology for OLED applications, exciplexes, namely emissive D–A intermolecular CT states, have been investigated in only very few computational studies.^{104,105} This is surely related to the weakness of intermolecular interactions in organics resulting in the large configurational space explored by the D and A molecules that can adopt multiple relative orientations. Besides sampling issues, another timely question relates to the nature of the quantum mechanical effect driving RISC in exciplexes, with scenarios based on either spin–orbit or hyperfine field couplings proposed in the literature.^{104,106} Note that recent studies of crystalline multichromophoric materials, composed on a mixed stack of 1:1 D–A molecules, are particularly interesting in this context, too.^{107,108}

As a last note, we stress that, as highlighted in this Perspective, modeling of TADF is a complex endeavor, and we warn the community about the potential pitfalls of "black box"

calculations using standard approaches. While these might provide a useful first screening, we believe that the field of computational modeling has now reached a maturity level that allows for a truly first-principles description of TADF emitters and hope this Perspective will guide modelers and experimentalists on finding their way to best practices in TADF.

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Notes

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ABBREVIATIONS

OLED: organic light emitting diode TADF: thermally activated delayed fluorescence IQE: Internal quantum efficiency T_n : *n*th triplet excited state S_n : *n*th singlet excited state (R)ISC: (reverse) intersystem crossing D: electron donating moiety/unit A: electron accepting moiety/unit CT: charge transfer LE: local excitation DFT: density functional theory TD-DFT: time-dependent density functional theory PCM: polarizable continuum models $\Delta E_{\rm ST}^{\rm V}$: vertical exchange gap/vertical singlet-triplet energy gap $\Omega(S_1)$: lowest singlet excited state energy $\Omega(T_1)$: lowest triplet excited state energy CCSDT: coupled-cluster with single, double, and (iterative) triple substitutions CC3: third-order approximate coupled-cluster EOM-CCSD: equation-of-motion coupled-cluster with single and double substitutions FCI: full configuration interactions CC2: second-order approximate coupled-cluster ADC(2): algebraic diagrammatic construction at second order ADC(3): algebraic diagrammatic construction at third order ADC(2)-x: extended algebraic diagrammatic construction at second order RAS: restricted active space CASSCF: complete active space self-consistent-field CASPT2: complete active space perturbation theory at second order GW+BSE: Green's functions with Bethe-Salpeter equation B3LYP: Becke three-parameter hybrid Lee-Yang-Parr exchange-correlation functional 6-31G^{*}: Pople's double- ξ basis set with polarization functions on second row atoms

M06-2X: Minnesota exchange–correlation functional in its 2006 version

TDA-DFT: time-dependent density functional theory in the Tamm-Dancoff approximation CIS: configuration interaction with single substitutions TD-HF: time-dependent Hartree-Fock PBE0: Perdew-Burke-Ernzerhof one-parameter hybrid exchange-correlation functional D3(BJ): dispersion correction (third-generation) with Becke-Johnson attenuation function def2-TZVP: Alhrichs' triple- ξ valence polarization basis set extended with diffuse function CAM-B3LYP: Coulomb-attenuating method B3LYP exchange-correlation functional ωB97X: ω-dependent range-separated Becke'97 exchangecorrelation functional GGA: Generalized Gradient Approximation HOMO: highest occupied molecular orbital LUMO: lowest-unoccupied molecular orbital UKS: unrestricted Kohn-Sham ROKS: restricted open-shell Kohn-Sham sTDA-DFT: simplified Tamm-Dancoff time-dependent density functional theory sTDA-xTB: simplified Tamm-Dancoff extended tightbinding Hamiltonian MP2: second-order Møller-Plesset perturbation method NPT: Isothermal-isobaric ensemble MRCI: multireference configuration interaction KS: Kohn-Sham orbitals NTO: natural transition orbitals $\Delta E_{\rm ST}$: singlet-triplet energy gap ZFS: zero-field splitting ESR: electron spin resonance 2CzPN: 4,5-di(9h-carbazol-9-yl)phthalonitrile 4CzIPN: 1,2,3,5-tetrakis(carbazol-9-yl)-4,6-dicyanobenzene (R)IC: (reverse) internal conversion ZORA: Zero-Order Relativistic Approximation MD: molecular dynamics MC: monte carlo QM: quantum mechanical NPT: Isothermal-isobaric ensemble CPU: central processing unit So: ground state ³CT: triplet charge-transfer state ³LE: triplet local excitation ε : dielectric constant TXO-TPA: 2-[4-(diphenylamino) phenyl]-10,10-dioxide-9H-thioxanthen-9-one TXO-PhCz: 2-(9-phenyl-9H-carbazol-3-yl)-10,10-dioxide-9H-thioxanthen-9-one

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