

Characterization of commercial iron oxide clusters as potential Magnetic Resonance Imaging contrast agent

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Commercial clusters of iron oxide particles, originally used for magnetofection, could constitute excellent MRI contrast agents

1. NMR relaxation induced by clusters of iron oxide particles

The efficiency of a contrast agent is given by its longitudinal and transverse relaxivities r_i

$$r_i = \left(\frac{1}{T_i} - \frac{1}{T_i^{dia}} \right) / [Fe] \quad \text{with } i=1,2 \quad \text{where} \quad \begin{cases} [Fe] \text{ is the iron concentration,} \\ 1/T_i \text{ is the relaxation rate of the solution,} \\ 1/T_i^{dia} \text{ is the relaxation rate of pure water,} \end{cases}$$

For magnetite particles with magnetization $M_v = 350\,000 \text{ A/m}$, the maximum r_2 is :

$$r_2^{max} \simeq 750 \text{ s}^{-1} \text{ mM}^{-1} \text{ that can be reached with } \begin{cases} \text{-large single crystals,} \\ \text{-large clusters of smaller iron oxide cores}^1, \end{cases}$$

For clusters of cores of given M_v , two parameters influence the relaxation regime:

(1) the radius R of the cluster and (2) the fraction ϕ_{intra} of the cluster volume occupied by cores

r_2^{max} is reached in the static dephasing regime (SDR) when $5 < \Delta\omega_{cluster} \tau_D^{cluster} < 20$

$$\text{with } \Delta\omega_{cluster} = \phi_{intra} \frac{\mu_0 \gamma M_v}{3} \quad \text{and} \quad \tau_D^{cluster} = \frac{R^2}{D} \quad [D = 3 \cdot 10^{-9} \text{ m}^2/\text{s} = \text{water diffusion coefficient,}$$

$$\gamma = 2,68 \cdot 10^8 \text{ Hz/T, } \mu_0 = 4\pi \cdot 10^{-7} \text{ Tm/A}]$$

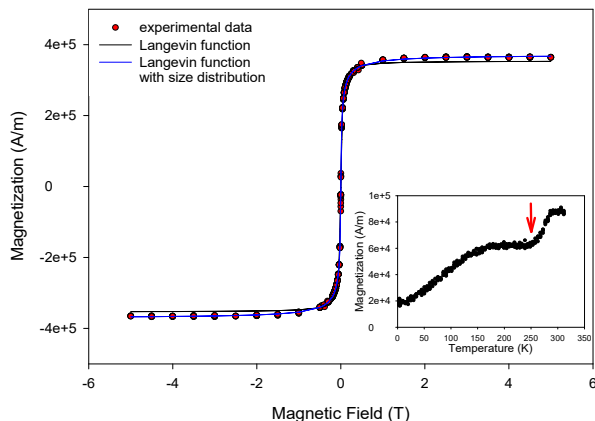


Figure 2: Effect of the field on the magnetization of the Polymag™ clusters. The inset shows the ZFC curve of the clusters.

2. NMR relaxation properties of the Polymag™ iron oxide clusters

Evolution with time of the transverse relaxation time T_2 when placed in the magnetic field,

⇒ reversible additional clustering in the field (disappears after removal from field and vortexing),

The evolution of $1/T_2$ with $[Fe]$ is shown in figure 4 for two magnetic fields (0.68 and 1.41 T),

$r_2 \sim 470 \text{ s}^{-1} \text{ mM}^{-1}$ is among the largest reported in literature, logical since $\Delta\omega_{cluster} \tau_D^{cluster} \sim 55$

⇒ The clusters are close to the SDR, without effect of temperature and interecho time on T_2

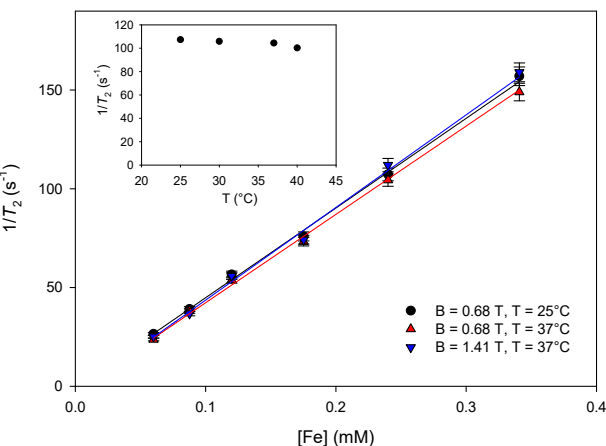


Figure 4: Evolution of $1/T_2$ with iron concentration of the Polymag™ aqueous solutions.

Table 1: relaxivities of the Polymag™ clusters.

	r_1 ($\text{s}^{-1} \text{ mM}^{-1}$)	r_2 ($\text{s}^{-1} \text{ mM}^{-1}$)
0.68 T, 25°C	20.5 ± 0.6	454 ± 12
0.68 T, 37°C	20.9 ± 0.7	447 ± 13
1.41 T, 37°C	9 ± 0.9	469 ± 17

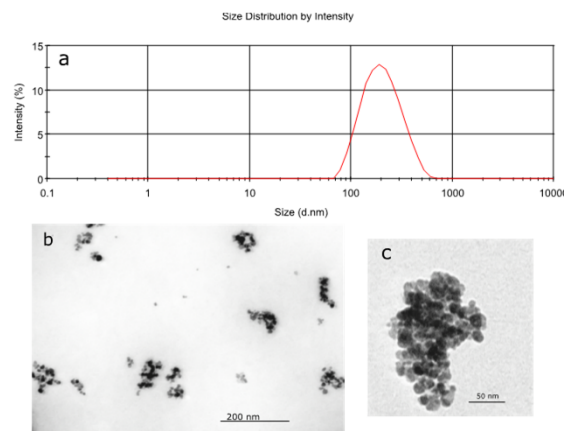


Figure 1: a) Hydrodynamic size distribution of the clusters obtained by DLS, b) TEM picture of several clusters, c) High magnification TEM image of a single cluster.

3. Characterization of the Polymag™ iron oxide clusters

Sample = Polymag™ clusters from Chemicell designed for magnetofection²,

Hydrodynamic size of the clusters estimated by Dynamic Light Scattering (DLS):

⇒ Z-average diameter = 180 nm (PDI = 0.15) - (fig.1a),

Images of the clusters obtained thanks to Transmission Electron Microscopy (TEM):

⇒ Dense clusters with cores of size comprised between 5 and 15 nm - (fig.1b and c),

Size distribution (assumed log-normal) of the cores obtained by magnetometry:

⇒ $d_{0,c} = 4.9 \pm 0.13 \text{ nm}$ and $\sigma = 0.53 \pm 0.01$ - (fig.2),

Zero field Cooling Curve (ZFC) with high blocking temperature and discontinuity at 273 K

⇒ typical of aqueous solutions of magnetic clusters (inset of fig.2),

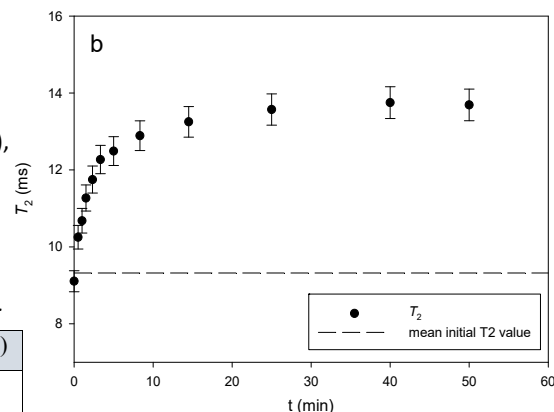


Figure 3: Evolution of T_2 with time after vortexing and insertion in the magnetic field at 0.68 T and 25°C.

5. Conclusions and perspectives

The ready-to-use Polymag™ clusters present excellent transverse relaxivity³ at 1.41 T,

The reversible clustering observed in the field could be a disadvantage,

Magnetofection could be used to load cells with the iron oxide clusters,

The relaxation properties of such loaded cells still has to be evaluated in vitro and in vivo

References

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 [2] Plank C, Zelpathi O and Mykhaylyk O 2011 Magnetically enhanced nucleic acid delivery. Ten years of magnetofection—Progress and prospects Advanced Drug Delivery Reviews 63 1300–31
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